

## ABSTRACT

*Water pollution by tartrazine is persistent and difficult to degrade. ZnO photocatalysis has the potential to address this problem, but it is limited to light response and electron-hole recombination. Mn doping is done to lower the band gap and increase activity in visible light. In the form of a membrane, glycerol is added as a plasticizer to improve its mechanical properties. This study aims to evaluate the photocatalytic performance of Mn-doped ZnO-based chitosan membranes in tartrazine degradation. This study aims to synthesize and evaluate the photocatalytic performance of Mn-doped ZnO-based chitosan membranes in tartrazine degradation under visible light. The membrane is synthesized by impregnating Mn-doped ZnO and ZnO into the chitosan matrix with and without the addition of glycerol. Characterization was performed using XRD, FTIR, SEM-EDX, and UV-Vis DRS. Photocatalytic activity was analyzed through tartrazine degradation assays using a first-order pseudo-kinetic model and statistically tested using ANOVA. The results showed that Mn-doped ZnO and ZnO had a wurtzite structure, with a decrease in crystal size from  $\pm 16$  nm to  $\pm 11.4$  nm and a decrease in band gap from  $\sim 3.2$  eV to  $\sim 2.7$  eV. The addition of glycerol increases the flexibility and hydrophilicity of the membrane, but decreases the distribution of active species on the surface. Tartrazine degradation followed first-order pseudo-kinetics ( $R^2 > 0.90$ ), with the highest rate constant on the 30% ZnO–Mn membrane without glycerol ( $k = 0.4432 \text{ min}^{-1}$ ). ANOVA analysis ( $p = 0.004$ ) showed significant differences between membrane variations. Chitosan–ZnO membranes doped with 30% Mn without glycerol exhibit the most optimal photocatalytic performance.*

**Keywords :** *Membrane, Mn-doped ZnO, Tartrazine, Photocatalytic Degradation.*