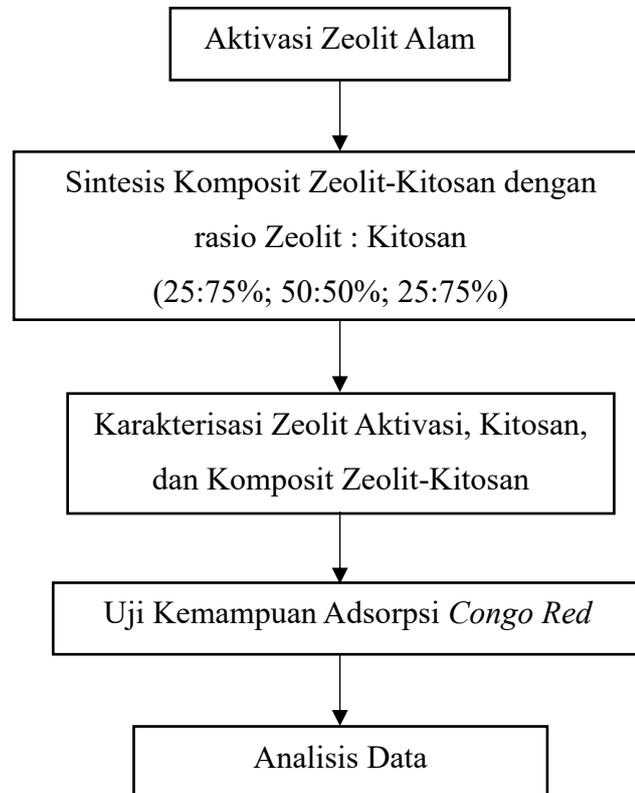
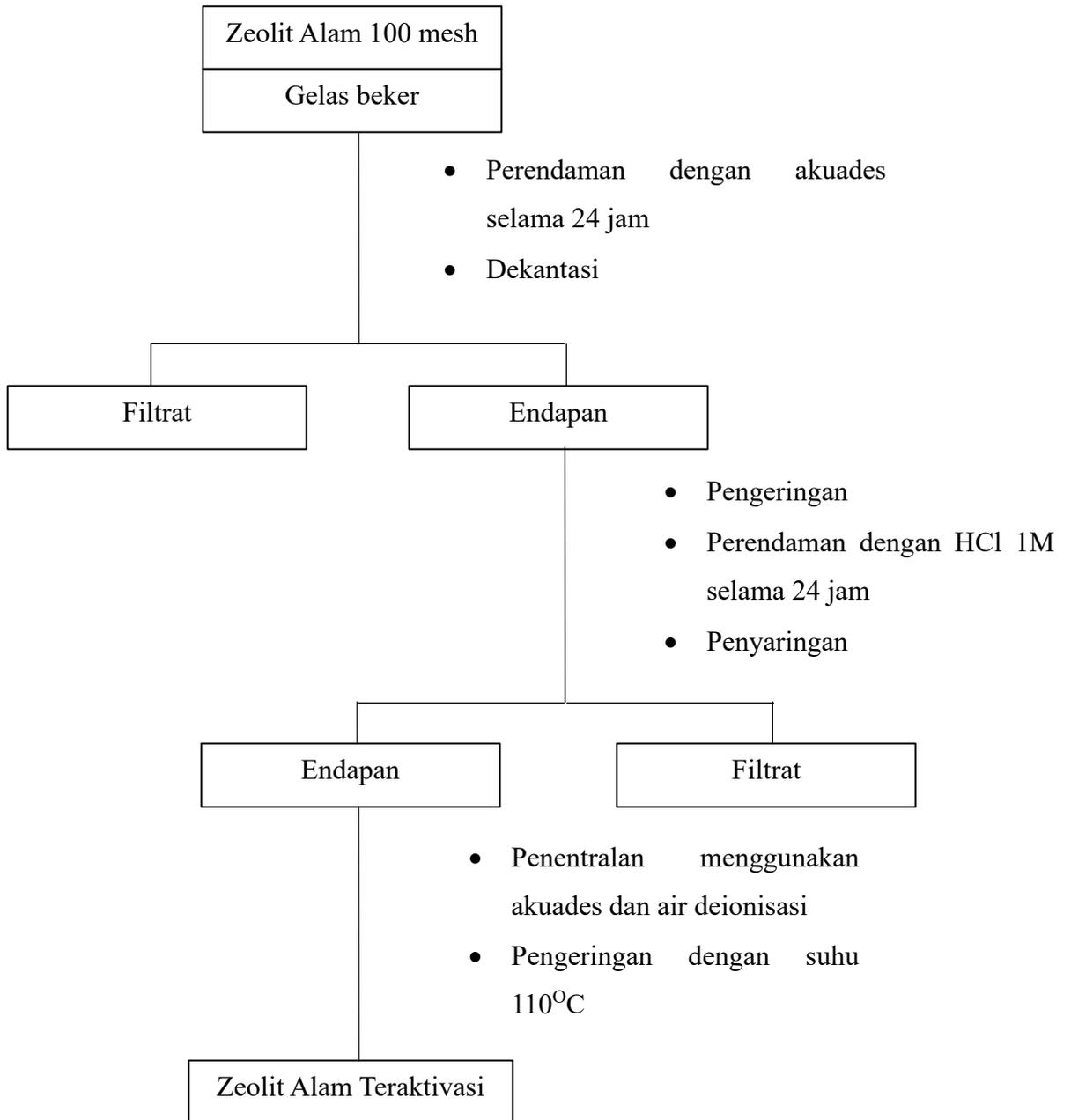


## LAMPIRAN

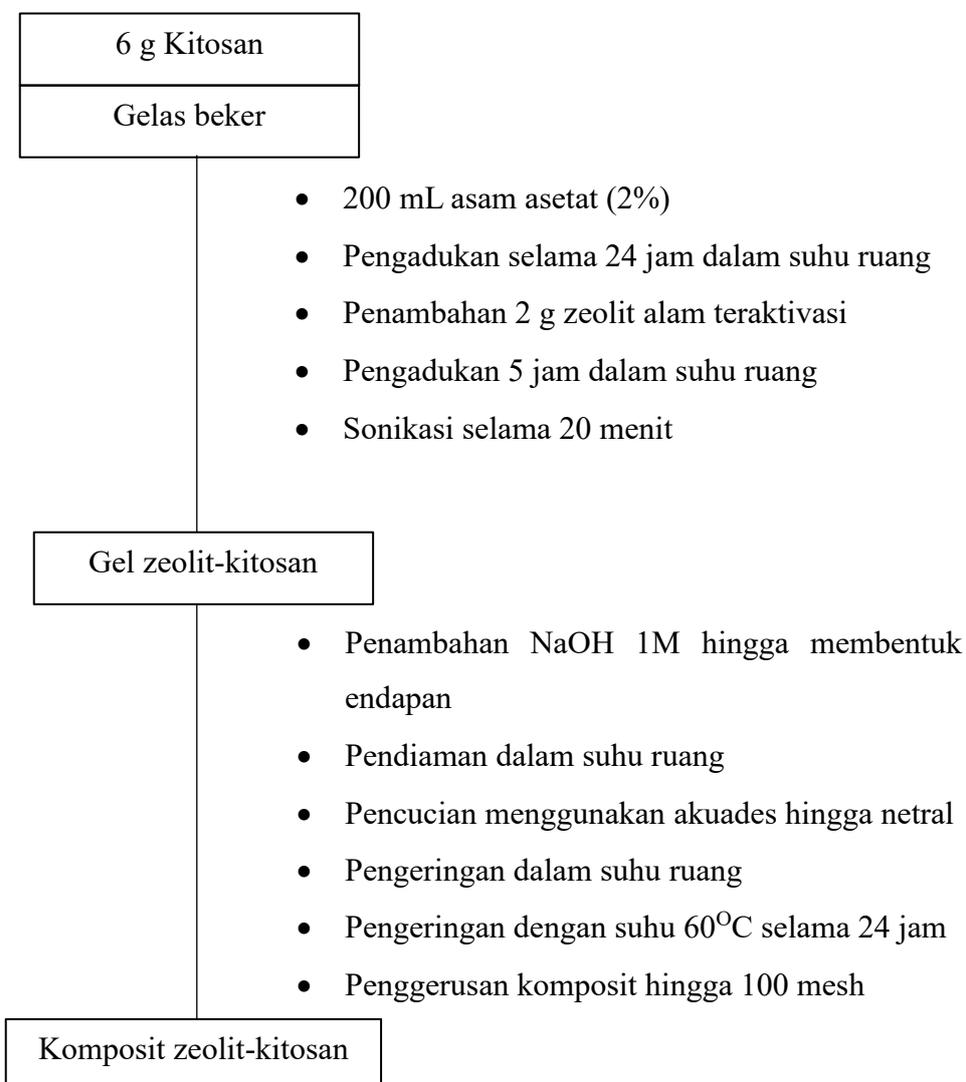
Lampiran 1. Diagram alur penelitian



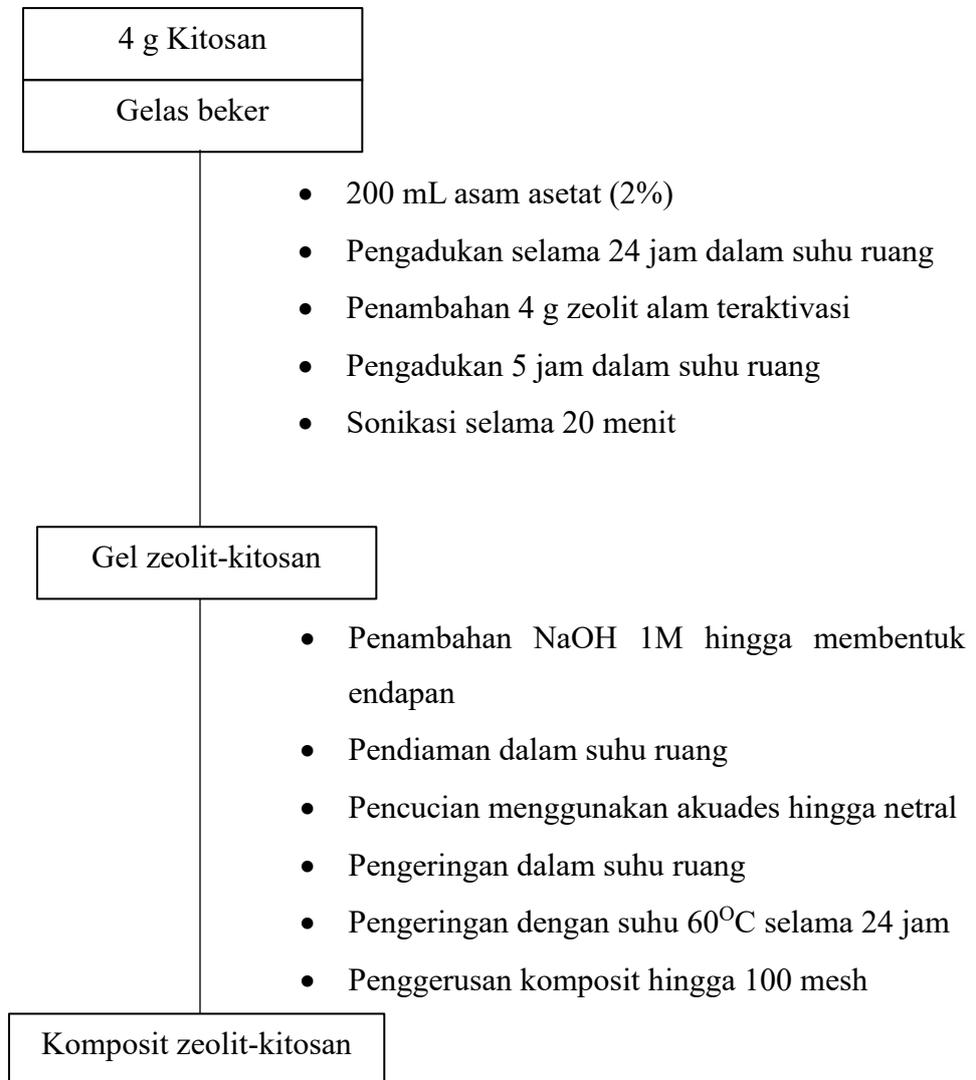
### 1.1 Aktivasi zeolit alam



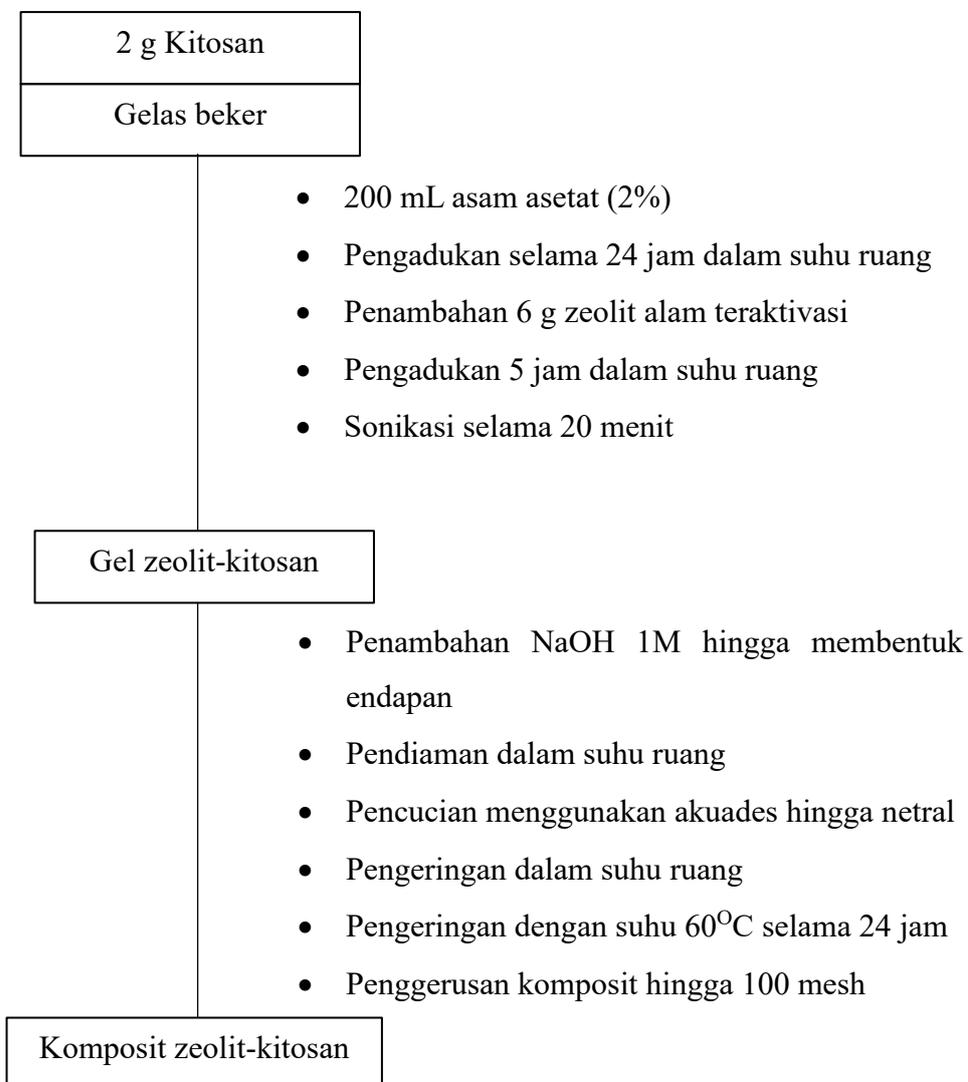
## 1.2 Sintesis komposit zeolit-kitosan (Z-Cs) 25%:75%



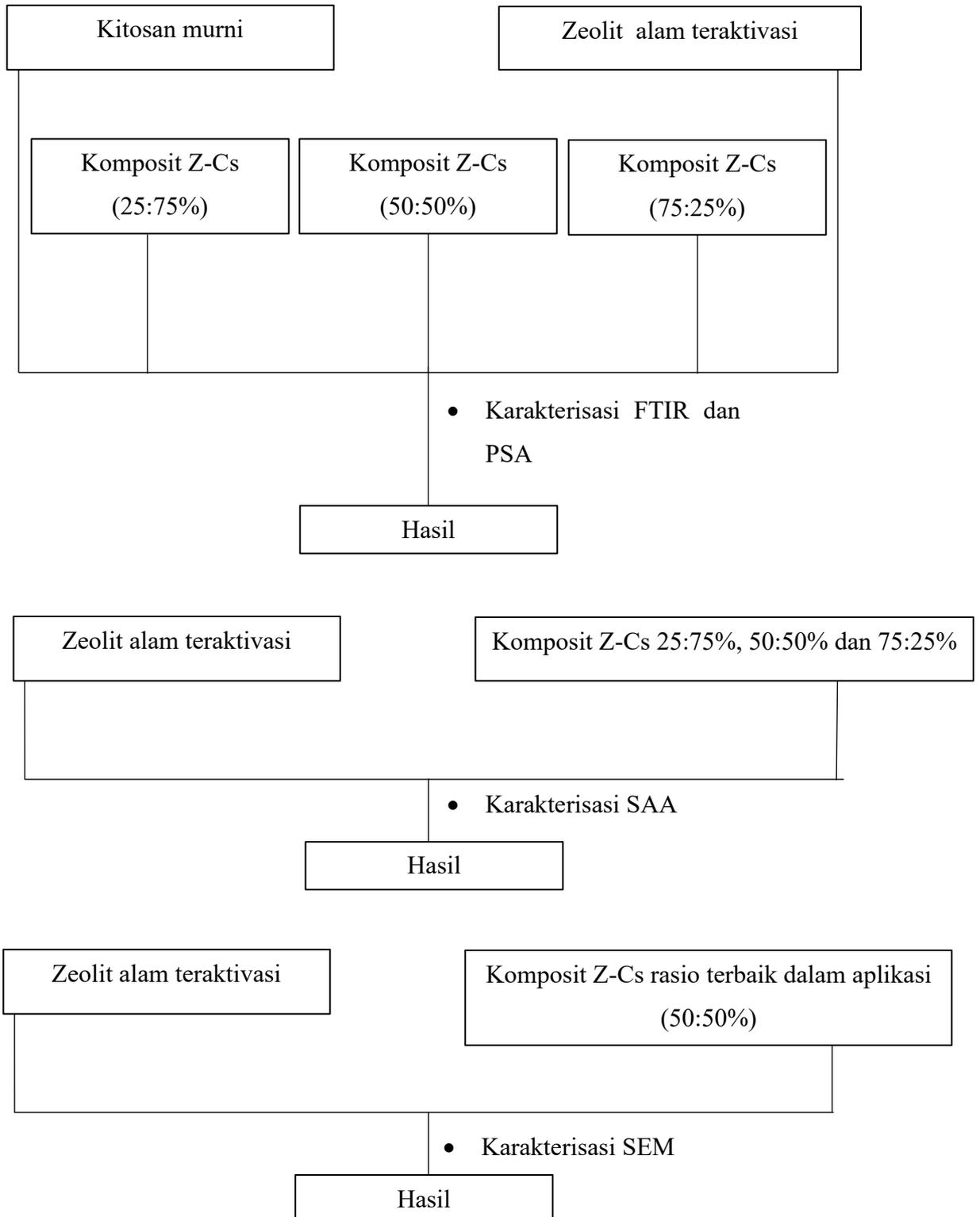
### 1.3 Sintesis komposit zeolit-kitosan (Z-Cs) 50%:50%



#### 1.4 Sintesis komposit zeolit-kitosan (Z-Cs) 75%:25%

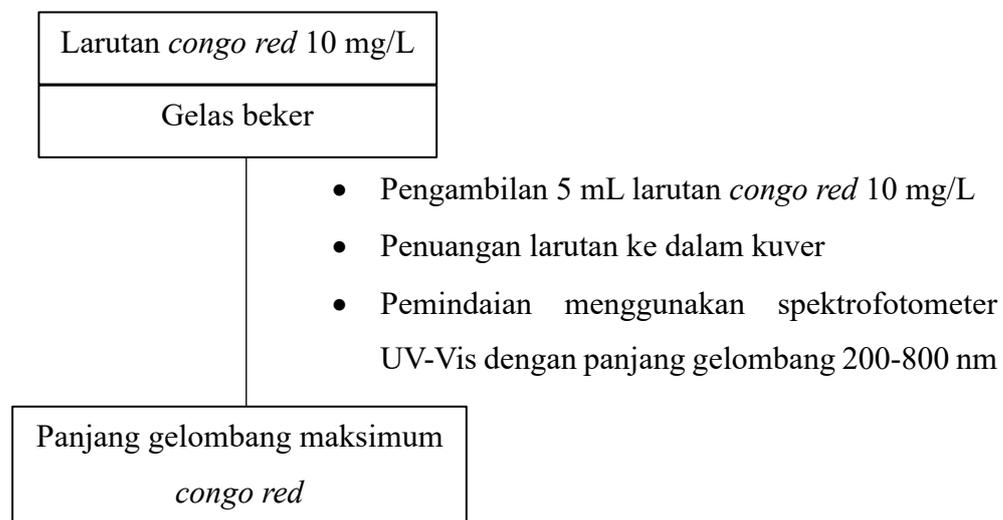


### 1.5 Karakterisasi material

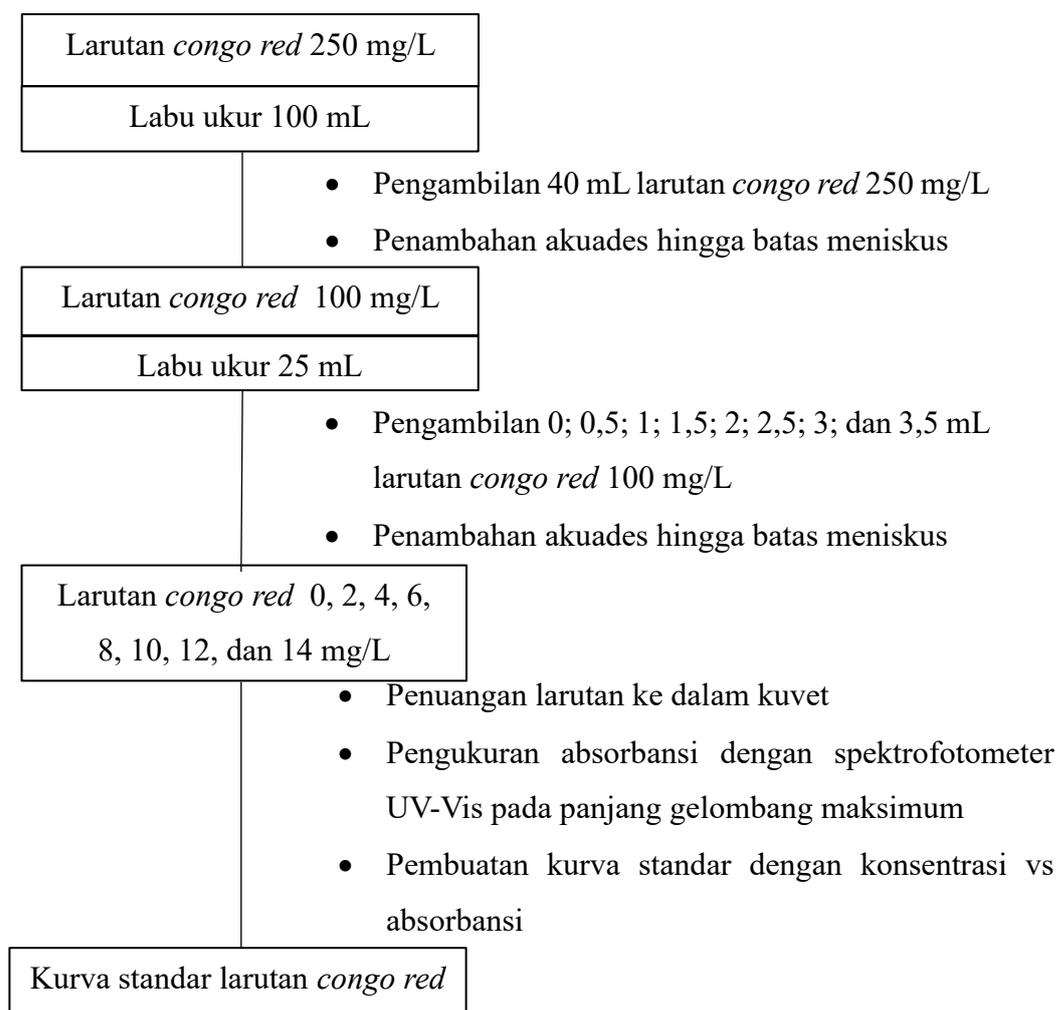


## 1.6 Penentuan panjang gelombang maksimum dan kurva standar *congo red*

### a. Penentuan Panjang Gelombang Maksimum *Congo Red*

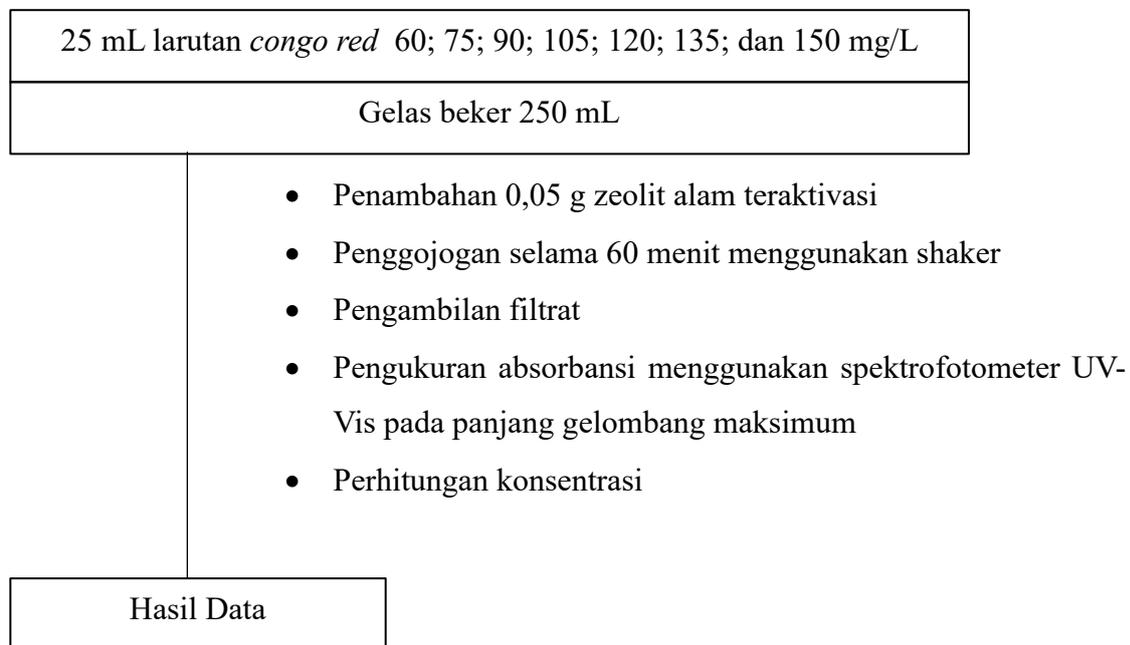


### b. Kurva Standar *Congo Red*

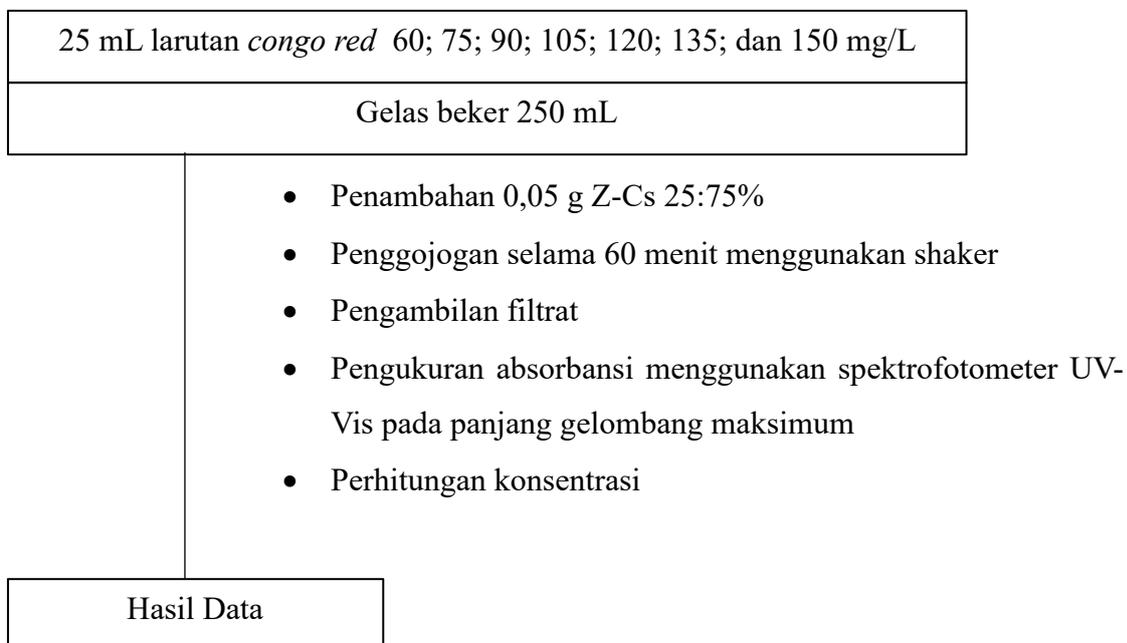


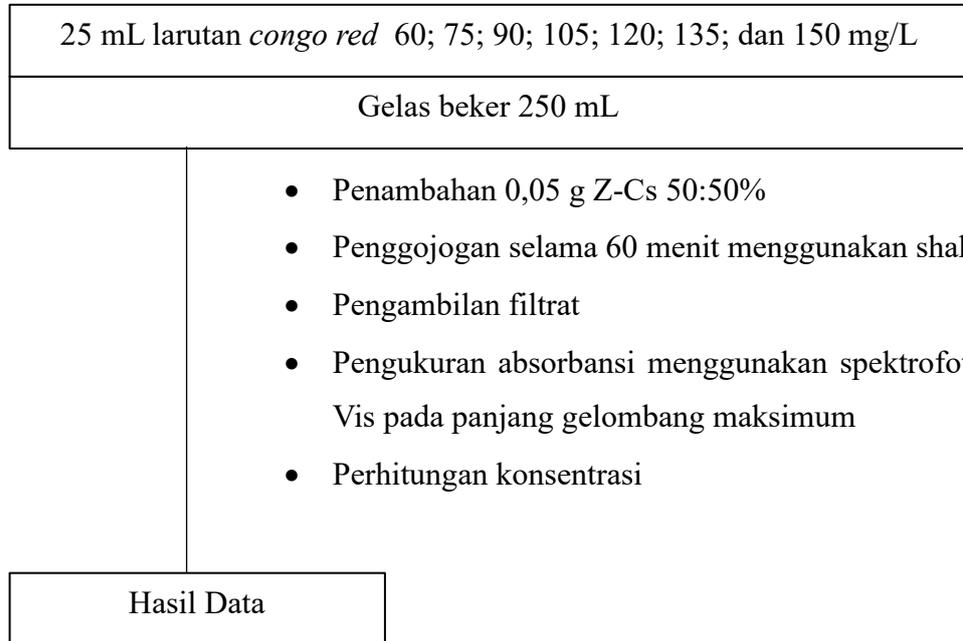
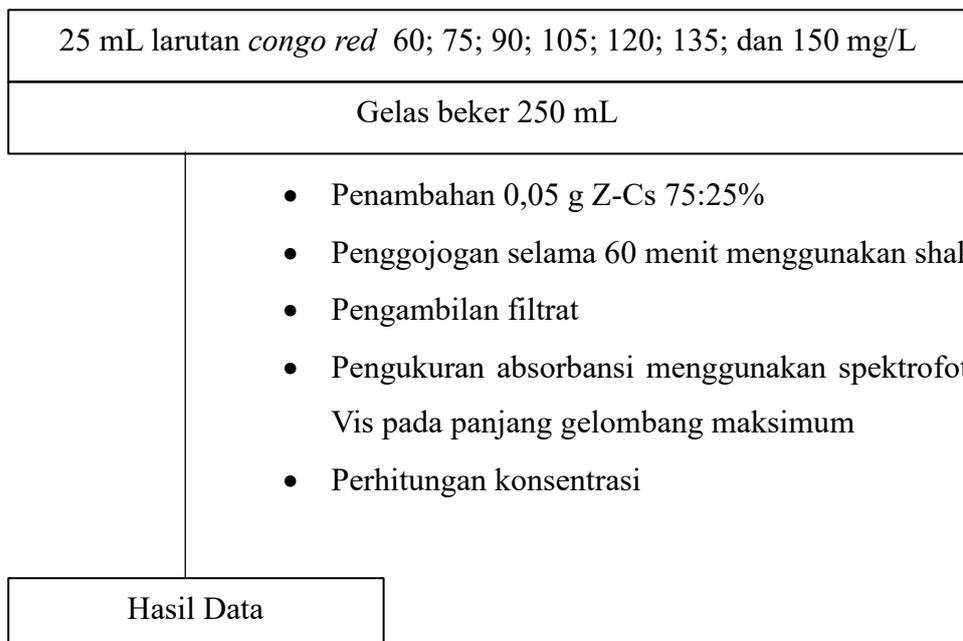
### 1.7 Pengujian adsorben pengaruh konsentrasi awal *congo red*

#### a. Zeolit alam teraktivasi



#### b. Z-Cs rasio 25:75%



**c. Z-Cs rasio 50:50%****d. Z-Cs rasio 75:25%**

## Lampiran 2. Perhitungan

### 2.1 Perhitungan aktivasi zeolit

Pembuatan HCl 1 M

$$M_{\text{HCl}} = \frac{\% \times \rho \times 1000}{M_r}$$

$$M_{\text{HCl}} = \frac{37 \% \times 1,19 \text{ g/mL} \times 1000}{36,5 \text{ g/mol}}$$

$$M_{\text{HCl}} = 12,06 \text{ mol/mL}$$

Untuk membuat HCl 1M 500 mL

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 12,06 \text{ M} = 500 \text{ mL} \times 1 \text{ M}$$

$$V_1 = 41,459 \text{ mL}$$

### 2.2 Perhitungan sintesis komposit zeolit-kitosan

a. Pembuatan 100 mL asam asetat 2% dari asam asetat 100%

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 100\% = 100 \text{ mL} \times 2\%$$

$$V_1 = 2 \text{ mL}$$

b. Pembuatan 100 mL NaOH 1M

$$M_{\text{NaOH}} = \frac{\text{mol}}{\text{L}}$$

$$\text{mol} = \frac{M_{\text{NaOH}}}{\text{L}} = \frac{1 \text{ M}}{0,1 \text{ L}} = 1 \text{ mol}$$

$$\text{mol} = \frac{\text{Massa}}{M_r}$$

$$1 \text{ mol} = \frac{\text{Massa}}{40 \text{ g/mol}}$$

$$\text{Massa} = 1 \text{ mol} \times 40 \text{ g/mol}$$

$$\text{Massa} = 4 \text{ g}$$

### 2.3 Perhitungan pembuatan larutan standar *congo red*

a. Pembuatan larutan induk 100 mL 250 mg/L *congo red*

$$250 \text{ mg/L} = 250 \text{ mg/L}$$

$$250 \text{ mg/L} = 250 \text{ mg}/1000 \text{ mL} = 0,25 \text{ mg/mL}$$

$$0,25 \text{ mg/mL} = x/100 \text{ mL}$$

$$x = 25 \text{ mg} = 0,025 \text{ gram}$$

b. Untuk 100 mL 100 mg/L *congo red* dari larutan induk

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 250 \text{ mg/L} = 100 \text{ mL} \times 100 \text{ mg/L}$$

$$V_1 = 40 \text{ mL}$$

c. Larutan standar 25 mL 0 mg/L *congo red* dari larutan 100 mg/L

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 100 \text{ mg/L} = 25 \text{ mL} \times 0 \text{ mg/L}$$

$$V_1 = 0 \text{ mL}$$

d. Larutan standar 25 mL 2 mg/L *congo red* dari larutan 100 mg/L

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 100 \text{ mg/L} = 25 \text{ mL} \times 2 \text{ mg/L}$$

$$V_1 = 0,5 \text{ mL}$$

e. Larutan standar 25 mL 4 mg/L *congo red* dari larutan 100 mg/L

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 100 \text{ mg/L} = 25 \text{ mL} \times 4 \text{ mg/L}$$

$$V_1 = 1 \text{ mL}$$

f. Larutan standar 25 mL 6 mg/L *congo red* dari larutan 100 mg/L

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 100 \text{ mg/L} = 25 \text{ mL} \times 6 \text{ mg/L}$$

$$V_1 = 1,5 \text{ mL}$$

g. Larutan standar 25 mL 8 mg/L *congo red* dari larutan 100 mg/L

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 100 \text{ mg/L} = 25 \text{ mL} \times 8 \text{ mg/L}$$

$$V_1 = 2 \text{ mL}$$

h. Larutan standar 25 mL 10 mg/L *congo red* dari larutan 100 mg/L

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 100 \text{ mg/L} = 25 \text{ mL} \times 10 \text{ mg/L}$$

$$V_1 = 2,5 \text{ mL}$$

i. Larutan standar 25 mL 12 mg/L *congo red* dari larutan 100 mg/L

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 100 \text{ mg/L} = 25 \text{ mL} \times 12 \text{ mg/L}$$

$$V_1 = 3 \text{ mL}$$

j. Larutan standar 25 mL 14 mg/L *congo red* dari larutan 100 mg/L

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 100 \text{ mg/L} = 25 \text{ mL} \times 14 \text{ mg/L}$$

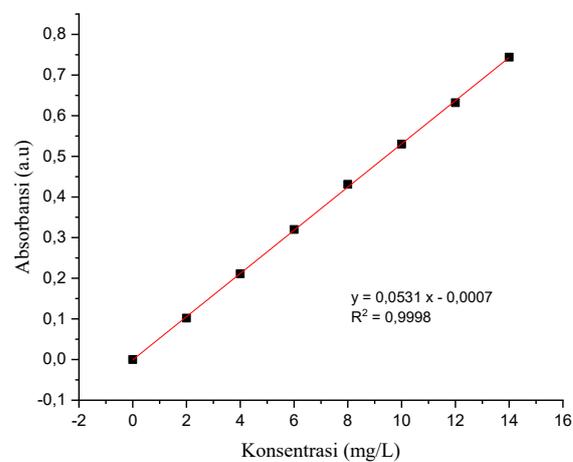
$$V_1 = 3,5 \text{ mL}$$

## 2.4 Perhitungan kurva standar *congo red*

a. Tabel

Kode Sampel	Konsentrasi (x)	Absorbansi (y)	xy	x <sup>2</sup>	y <sup>2</sup>
0 mg/L	0	0	0	0	0
2 mg/L	2	0,102	0,204	4	0,010404
4 mg/L	4	0,211	0,844	16	0,044521
6 mg/L	6	0,320	1,920	36	0,1024
8 mg/L	8	0,431	3,448	64	0,185761
10 mg/L	10	0,530	5,300	100	0,2809
12 mg/L	12	0,632	7,584	144	0,399424
14 mg/L	14	0,744	10,416	196	0,553536
Jumlah	56	2,97	29,716	560	1,576946
Rata-rata	7	0,37125	3,7145	70	0,197118

b. Grafik



## c. Persamaan Garis

1. Mencari  $m$ 

$$m = \frac{n (\sum xy) - (\sum x)(\sum y)}{n (\sum x^2) - (\sum x)^2}$$

$$m = \frac{8 (29,716) - (56)(2,97)}{8 (560) - (\sum 56)^2}$$

$$m = 0,0531$$

Dengan  $n = 8$

2. Mencari  $c$ 

$$\bar{y} = m \bar{x} + c$$

$$c = \bar{y} - m \bar{x}$$

$$c = 0,37125 - 0,05313 \times 7$$

$$c = -0,0007$$

3. Mencari  $R^2$ 

$$R^2 = \frac{[n (\sum xy) - (\sum x)(\sum y)]^2}{[n(\sum x^2) - (\sum x)^2] [n(\sum y^2) - (\sum y)^2]}$$

$$R^2 = \frac{[8 (29,716) - (56)(2,97)]^2}{[8(560) - (56^2)][8(1,576946) - (2,97^2)]}$$

$$R^2 = 0,9998$$

Maka persamaan garis yang didapatkan

$$\bar{y} = 0,0531 \bar{x} + 0,0007$$

$$R^2 = 0,9998$$

**2.5** Pembuatan variasi konsentrasi larutan *congo red* sebagai adsorbata. Larutan *congo red* 100 mL 250 mg/L

$$250 \text{ mg/L} = 250 \text{ mg/L}$$

$$250 \text{ mg/L} = 250 \text{ mg}/1000 \text{ mL} = 0,25 \text{ mg/mL}$$

$$0,25 \text{ mg/mL} = x/100 \text{ mL}$$

$$x = 25 \text{ mg} = 0,025 \text{ gram}$$

## b. Variasi 60 mg/L dengan labu ukur 25 mL

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 250 \text{ mg/L} = 25 \text{ mL} \times 60 \text{ mg/L}$$

$$V_1 = 6 \text{ mL}$$

c. Variasi 75 mg/L dengan labu ukur 25 mL

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 250 \text{ mg/L} = 25 \text{ mL} \times 75 \text{ mg/L}$$

$$V_1 = 7,5 \text{ mL}$$

d. Variasi 90 mg/L dengan labu ukur 25 mL

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 250 \text{ mg/L} = 25 \text{ mL} \times 90 \text{ mg/L}$$

$$V_1 = 9 \text{ mL}$$

e. Variasi 105 mg/L dengan labu ukur 25 mL

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 250 \text{ mg/L} = 25 \text{ mL} \times 105 \text{ mg/L}$$

$$V_1 = 10,5 \text{ mL}$$

f. Variasi 120 mg/L dengan labu ukur 25 mL

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 250 \text{ mg/L} = 25 \text{ mL} \times 120 \text{ mg/L}$$

$$V_1 = 12 \text{ mL}$$

g. Variasi 135 mg/L dengan labu ukur 25 mL

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 250 \text{ mg/L} = 25 \text{ mL} \times 135 \text{ mg/L}$$

$$V_1 = 13,5 \text{ mL}$$

h. Variasi 150 mg/L dengan labu ukur 25 mL

$$V_1 \times M_1 = V_2 \times M_2$$

$$V_1 \times 250 \text{ mg/L} = 25 \text{ mL} \times 150 \text{ mg/L}$$

$$V_1 = 15 \text{ mL}$$

## 2.6 Perhitungan konsentrasi setelah adsorpsi ( $C_e$ ) *congo red*

$$y = mx + c$$

$$y = 0,0531 x - 0,0007$$

$$x = \frac{y - c}{m}$$

$$C_e = x \times FP$$

Keterangan :

$y$  = absorbansi

$x$  = konsentrasi akhir ( $\frac{\text{mg}}{\text{L}}$ )

$C_e$  = konsentrasi setelah adsorpsi ( $\frac{\text{mg}}{\text{L}}$ )

$FP$  = Faktor Pengenceran

$$c = -0,0007$$

$$m = 0,0531$$

a. Adsorben Zeolit Alam Teraktivasi

$$60 \text{ mg/L}, y = 0,469$$

Dengan  $FP = 5$

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,469 - (-0,0007)}{0,0531} \times 5$$

$$C_e = 44,2279 \text{ mg/L}$$

$$75 \text{ mg/L}, y = 0,609$$

Dengan  $FP = 5$

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,609 - (-0,0007)}{0,0531} \times 5$$

$$C_e = 57,4105 \text{ mg/L}$$

$$90 \text{ mg/L}, y = 0,377$$

Dengan  $FP = 10$

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,377 - (-0,0007)}{0,0531} \times 10$$

$$C_e = 71,1299 \text{ mg/L}$$

$$105 \text{ mg/L}, y = 0,444$$

Dengan  $FP = 10$

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,444 - (-0,0007)}{0,0531} \times 10$$

$$C_e = 83,7476 \text{ mg/L}$$

$$120 \text{ mg/L}, y = 0,513$$

dengan  $FP = 10$

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,513 - (-0,0007)}{0,0531} \times 10$$

$$C_e = 96,742 \text{ mg/L}$$

$$135 \text{ mg/L}, y = 0,569$$

dengan  $FP = 10$

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,569 - (-0,0007)}{0,0531} \times 10$$

$$C_e = 107,2881 \text{ mg/L}$$

150 mg/L,  $y = 0,617$

dengan  $FP = 10$

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,617 - (-0,0007)}{0,0531} \times 10$$

$$C_e = 116,3277 \text{ mg/L}$$

b. Adsorben Z-Cs 25:75%

60 mg/L,  $y = 0,054$

$$C_e = \frac{y - c}{m}$$

$$C_e = \frac{0,054 - (-0,0007)}{0,0531}$$

$$C_e = 1,0301 \text{ mg/L}$$

75 mg/L,  $y = 0,107$

$$C_e = \frac{y - c}{m}$$

$$C_e = \frac{0,107 - (-0,0007)}{0,0531}$$

$$C_e = 2,0282 \text{ mg/L}$$

90 mg/L,  $y = 0,394$

$$C_e = \frac{y - c}{m}$$

$$C_e = \frac{0,394 - (-0,0007)}{0,0531}$$

$$C_e = 7,4331 \text{ mg/L}$$

105 mg/L,  $y = 0,769$

$$C_e = \frac{y - c}{m}$$

$$C_e = \frac{0,769 - (-0,0007)}{0,0531}$$

$$C_e = 14,4953 \text{ mg/L}$$

120 mg/L,  $y = 0,278$

dengan  $FP = 5$

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,278 - (-0,0007)}{0,0531} \times 5$$

$$C_e = 26,2429 \text{ mg/L}$$

135 mg/L,  $y = 0,326$

dengan  $FP = 5$

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,326 - (-0,0007)}{0,0531} \times 5$$

$$C_e = 30,7627 \text{ mg/L}$$

150 mg/L,  $y = 0,402$

dengan  $FP = 5$

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,402 - (-0,0007)}{0,0531} \times 5$$

$$C_e = 37,9190 \text{ mg/L}$$

c. Adsorben Z-Cs 50:50%

$$60 \text{ mg/L, } y = 0,014$$

$$C_e = \frac{y - c}{m}$$

$$C_e = \frac{0,014 - (-0,0007)}{0,0531}$$

$$C_e = 0,2768 \text{ mg/L}$$

$$75 \text{ mg/L, } y = 0,033$$

$$C_e = \frac{y - c}{m}$$

$$C_e = \frac{0,033 - (-0,0007)}{0,0531}$$

$$C_e = 0,6347 \text{ mg/L}$$

$$90 \text{ mg/L, } y = 0,108$$

$$C_e = \frac{y - c}{m}$$

$$C_e = \frac{0,108 - (-0,0007)}{0,0531}$$

$$C_e = 2,0471 \text{ mg/L}$$

$$105 \text{ mg/L, } y = 0,321$$

$$C_e = \frac{y - c}{m}$$

$$C_e = \frac{0,321 - (-0,0007)}{0,0531}$$

$$C_e = 6,0584 \text{ mg/L}$$

$$120 \text{ mg/L, } y = 0,357$$

Dengan FP = 2

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,357 - (-0,0007)}{0,0531} \times 2$$

$$C_e = 13,4727 \text{ mg/L}$$

$$135 \text{ mg/L, } y = 0,222$$

Dengan FP = 5

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,222 - (-0,0007)}{0,0531} \times 5$$

$$C_e = 20,9699 \text{ mg/L}$$

$$150 \text{ mg/L, } y = 0,311$$

Dengan FP = 5

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,311 - (-0,0007)}{0,0531} \times 5$$

$$C_e = 29,3503 \text{ mg/L}$$

d. Adsorben Z-Cs 75:25%

$$60 \text{ mg/L, } y = 0,402$$

$$C_e = \frac{y - c}{m}$$

$$C_e = \frac{0,402 - (-0,0007)}{0,0531}$$

$$C_e = 7,5838 \text{ mg/L}$$

$$75 \text{ mg/L, } y = 0,755$$

$$C_e = \frac{y - c}{m}$$

$$C_e = \frac{0,609 - (-0,0007)}{0,0531}$$

$$C_e = 14,2316 \text{ mg/L}$$

$$90 \text{ mg/L, } y = 0,519$$

Dengan FP = 2

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,519 - (-0,0007)}{0,0531} \times 2$$

$$C_e = 19,5744 \text{ mg/L}$$

$$105 \text{ mg/L, } y = 0,302$$

Dengan FP = 5

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,302 - (-0,0007)}{0,0531} \times 5$$

$$C_e = 28,5028 \text{ mg/L}$$

$$120 \text{ mg/L, } y = 0,428$$

dengan FP = 5

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,428 - (-0,0007)}{0,0531} \times 5$$

$$C_e = 40,3672 \text{ mg/L}$$

$$135 \text{ mg/L, } y = 0,243$$

dengan FP = 10

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,243 - (-0,0007)}{0,0531} \times 10$$

$$C_e = 45,8945 \text{ mg/L}$$

$$150 \text{ mg/L, } y = 0,277$$

dengan FP = 10

$$C_e = \frac{y - c}{m} \times FP$$

$$C_e = \frac{0,277 - (-0,0007)}{0,0531} \times 10$$

$$C_e = 52,2976 \text{ mg/L}$$

## 2.7 Perhitungan kemampuan adsorpsi ( $q_e$ )

$$q_e = \frac{(C_i - C_e) \times V}{W}$$

Keterangan :

$C_i$  = konsentrasi awal ( $\frac{\text{mg}}{\text{L}}$ )

$C_e$  = konsentrasi setelah adsorpsi ( $\frac{\text{mg}}{\text{L}}$ )

$V$  = Volume adsorbat (L) = 0,025 L

$W$  = massa adsorben (g) = 0,05 g

### a. Adsorben Zeolit Alam Teraktivasi

$$C_i = 60 \frac{\text{mg}}{\text{L}}$$

$$C_e = 44,2279 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(60 - 44,2279) \text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 7,8861 \text{ mg/g}$$

$$C_i = 75 \frac{\text{mg}}{\text{L}}$$

$$C_e = 57,4105 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(75 - 57,4105) \text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 8,7947 \text{ mg/g}$$

$$C_i = 90 \frac{\text{mg}}{\text{L}}$$

$$C_e = 71,1299 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(90 - 71,1299) \text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 9,4350 \text{ mg/g}$$

$$C_i = 105 \frac{\text{mg}}{\text{L}}$$

$$C_e = 83,7476 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(105 - 83,7476) \text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 10,6262 \text{ mg/g}$$

$$C_i = 120 \frac{\text{mg}}{\text{L}}$$

$$C_e = 96,7420 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(120 - 96,7420) \text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 11,6290 \text{ mg/g}$$

$$C_i = 135 \frac{\text{mg}}{\text{L}}$$

$$C_e = 107,2881 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(135 - 107,2881) \text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 13,8559 \text{ mg/g}$$

$$C_i = 150 \frac{\text{mg}}{\text{L}}$$

$$C_e = 116,3277 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(150-116,3277)\text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 16,8362 \text{ mg/g}$$

b. Adsorben Z-Cs 25:75%

$$C_i = 60 \frac{\text{mg}}{\text{L}}$$

$$C_e = 1,0301 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(60-1,0301)\text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 29,4849 \text{ mg/g}$$

$$C_i = 75 \frac{\text{mg}}{\text{L}}$$

$$C_e = 2,0282 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(75-2,0282)\text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 36,4859 \text{ mg/g}$$

$$C_i = 90 \frac{\text{mg}}{\text{L}}$$

$$C_e = 7,4331 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(90 - 7,4331)\text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 41,2834 \text{ mg/g}$$

$$C_i = 105 \frac{\text{mg}}{\text{L}}$$

$$C_e = 14,4953 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(105-14,4953)\text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 45,2524 \text{ mg/g}$$

$$C_i = 120 \frac{\text{mg}}{\text{L}}$$

$$C_e = 26,2429 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(120-26,2429)\text{mg}}{\text{L}} \cdot 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 46,8785 \text{ mg/g}$$

$$C_i = 135 \frac{\text{mg}}{\text{L}}$$

$$C_e = 30,7627 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(135-30,7627)\text{mg}}{\text{L}} \cdot 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 52,1186 \text{ mg/g}$$

$$C_i = 150 \frac{\text{mg}}{\text{L}}$$

$$C_e = 37,9190 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(150-37,9190)\text{mg}}{\text{L}} \cdot 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 56,0405 \text{ mg/g}$$

c. Adsorben Z-Cs 50:50%

$$C_i = 60 \frac{\text{mg}}{\text{L}}$$

$$C_e = 0,2768 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(60-0,2768)\text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 29,8616 \text{ mg/g}$$

$$C_i = 90 \frac{\text{mg}}{\text{L}}$$

$$C_e = 2,0471 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(90-2,0471)\text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 43,9765 \text{ mg/g}$$

$$C_i = 120 \frac{\text{mg}}{\text{L}}$$

$$C_e = 13,4727 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(120-13,4727)\text{mg}}{\text{L}} \cdot 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 53,2637/\text{g}$$

$$C_i = 150 \frac{\text{mg}}{\text{L}}$$

$$C_e = 29,3503 \frac{\text{mg}}{\text{L}}$$

$$C_i = 75 \frac{\text{mg}}{\text{L}}$$

$$C_e = 0,6347 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(75-0,6347)\text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 37,1827 \text{ mg/g}$$

$$C_i = 105 \frac{\text{mg}}{\text{L}}$$

$$C_e = 6,0584 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(105-6,0584)\text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 49,4708 \text{ mg/g}$$

$$C_i = 135 \frac{\text{mg}}{\text{L}}$$

$$C_e = 20,9699 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(135-20,9699)\text{mg}}{\text{L}} \cdot 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 57,0151 \text{ mg/g}$$

$$q_e = \frac{\frac{(120-29,3503)\text{mg}}{\text{L}} \cdot 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 60,3249 \text{ mg/g}$$

d. Adsorben Z-Cs 75:25%

$$C_i = 60 \frac{\text{mg}}{\text{L}}$$

$$C_e = 7,5838 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(60-7,5838)\text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 26,2081 \text{ mg/g}$$

$$C_i = 90 \frac{\text{mg}}{\text{L}}$$

$$C_e = 19,5744 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(90-19,5744)\text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 35,2128 \text{ mg/g}$$

$$C_i = 120 \frac{\text{mg}}{\text{L}}$$

$$C_e = 40,3672 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(120-40,3672)\text{mg}}{\text{L}} \cdot 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 39,8164 \text{ mg/g}$$

$$C_i = 150 \frac{\text{mg}}{\text{L}}$$

$$C_e = 52,2976 \frac{\text{mg}}{\text{L}}$$

$$C_i = 75 \frac{\text{mg}}{\text{L}}$$

$$C_e = 14,2316 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(75-14,2316)\text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 30,3842 \text{ mg/g}$$

$$C_i = 105 \frac{\text{mg}}{\text{L}}$$

$$C_e = 28,5028 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(105-28,5028)\text{mg}}{\text{L}} \times 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 38,2486 \text{ mg/g}$$

$$C_i = 135 \frac{\text{mg}}{\text{L}}$$

$$C_e = 45,8945 \frac{\text{mg}}{\text{L}}$$

$$q_e = \frac{\frac{(135-45,8945)\text{mg}}{\text{L}} \cdot 0,025 \text{ L}}{0,05 \text{ g}}$$

$$q_e = 44,5527 \text{ mg/g}$$

$$q_e = \frac{(150 - 52,2976) \text{ mg}}{L} \cdot 0,025 \text{ L}$$

$$q_e = \frac{\quad}{0,05 \text{ g}}$$

$$q_e = 48,8512 \text{ mg/g}$$

### Lampiran 3. Tabel dan Grafik

#### 3.1 Tabel kemampuan dan efisiensi adsorpsi

##### a. Adsorben Zeolit Alam Teraktivasi

Ci (mg/L)	y	m	c	FP	Ce (mg/L)	C teradsorpsi (mg/L)	qe (mg/g)	%Efisiensi
60	0,469	0,0531	-0,0007	5	44,2279	15,7721	7,8861	26,2869
75	0,609	0,0531	-0,0007	5	57,4105	17,5895	8,7947	23,4526
90	0,377	0,0531	-0,0007	10	71,1299	18,8701	9,4350	20,9667
105	0,444	0,0531	-0,0007	10	83,7476	21,2524	10,6262	20,2403
120	0,513	0,0531	-0,0007	10	96,7420	23,2580	11,6290	19,3817
135	0,569	0,0531	-0,0007	10	107,2881	27,7119	13,8559	20,5273
150	0,617	0,0531	-0,0007	10	116,3277	33,6723	16,8362	22,4482

##### b. Adsorben Z-Cs 25:75%

Ci (mg/L)	y	m	c	FP	Ce (mg/L)	C teradsorpsi (mg/L)	qe (mg/g)	%Efisiensi
60	0,054	0,0531	-0,0007	1	1,0301	58,9699	29,4849	98,2831
75	0,107	0,0531	-0,0007	1	2,0282	72,9718	36,4859	97,2957
90	0,394	0,0531	-0,0007	1	7,4331	82,5669	41,2834	91,7409
105	0,769	0,0531	-0,0007	1	14,4953	90,5047	45,2524	86,1950
120	0,278	0,0531	-0,0007	5	26,2429	93,7571	46,8785	78,1309
135	0,326	0,0531	-0,0007	5	30,7627	104,2373	52,1186	77,2128
150	0,402	0,0531	-0,0007	5	37,9190	112,0810	56,0405	74,7207

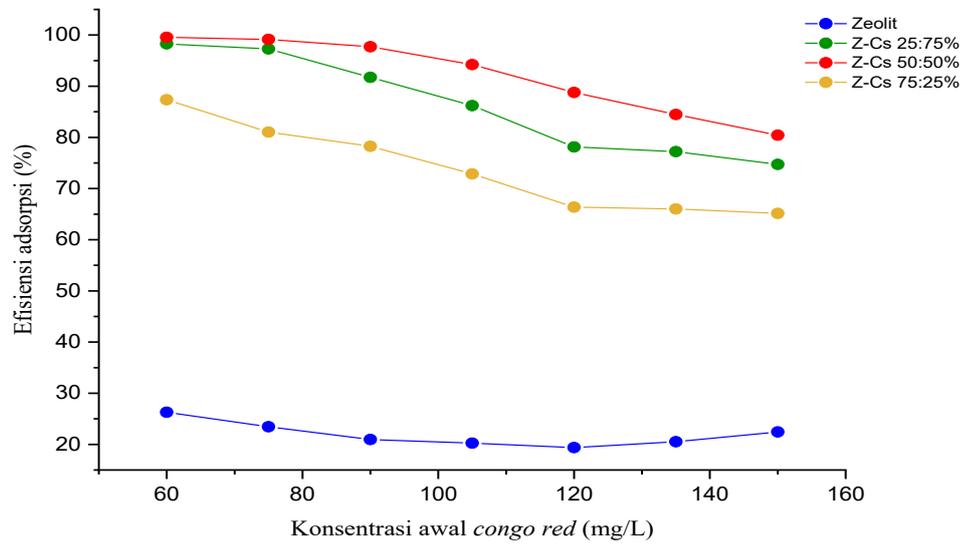
## c. Adsorben Z-Cs 50:50%

Ci (mg/L)	y	m	c	FP	Ce (mg/L)	C teradsorpsi (mg/L)	qe (mg/g)	%Efisiensi
60	0,014	0,0531	-0,0007	1	0,2768	59,7232	29,8616	99,5386
75	0,033	0,0531	-0,0007	1	0,6347	74,3653	37,1827	99,1538
90	0,108	0,0531	-0,0007	1	2,0471	87,9529	43,9765	97,7255
105	0,321	0,0531	-0,0007	1	6,0584	98,9416	49,4708	94,2301
120	0,357	0,0531	-0,0007	2	13,4727	106,5273	53,2637	88,7728
135	0,222	0,0531	-0,0007	5	20,9699	114,0301	57,0151	84,4668
150	0,311	0,0531	-0,0007	5	29,3503	120,6497	60,3249	80,4331

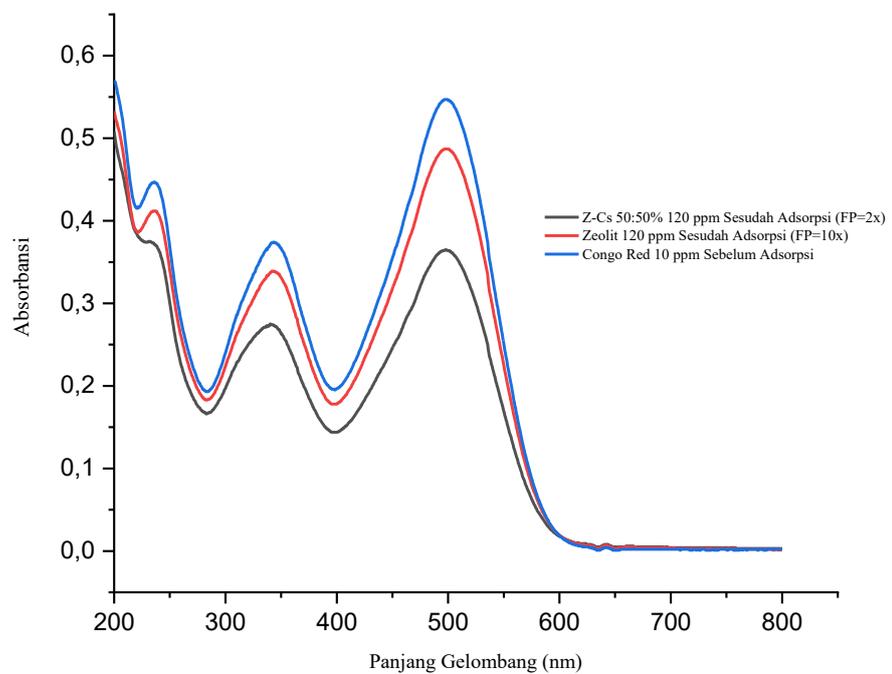
## d. Adsorben Z-Cs 75:25%

Ci (mg/L)	y	m	c	FP	Ce (mg/L)	C teradsorpsi (mg/L)	qe (mg/g)	%Efisiensi
60	0,402	0,0531	-0,0007	1	7,5838	52,4162	26,2081	87,3603
75	0,755	0,0531	-0,0007	1	14,2316	60,7684	30,3842	81,0245
90	0,519	0,0531	-0,0007	2	19,5744	70,4256	35,2128	78,2507
105	0,302	0,0531	-0,0007	5	28,5028	76,4972	38,2486	72,8545
120	0,428	0,0531	-0,0007	5	40,3672	79,6328	39,8164	66,3606
135	0,243	0,0531	-0,0007	10	45,8945	89,1055	44,5527	66,0040
150	0,277	0,0531	-0,0007	10	52,2976	97,7024	48,8512	65,1350

### 3.2 Grafik efisiensi adsorpsi (%)

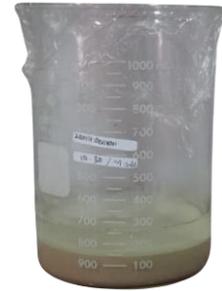


### 3.3 Spektrum *congo red* sebelum dan sesudah adsorpsi

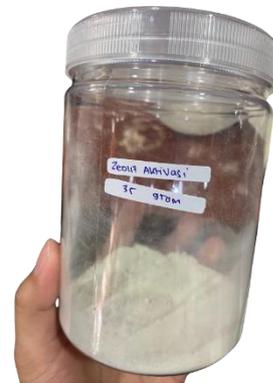


**Lampiran 4. Dokumentasi penelitian**

Zeolit Alam



Aktivasi Zeolit menggunakan HCl

Pencucian Zeolit Aktivasi menggunakan  
akuades hingga netral

Zeolit teraktivasi asam



Pelarutan Kitosan



Suspensi Zeolit-Kitosan



Sonikasi Suspensi Zeolit-Kitosan



Penambahan NaOH



Komposit Z-Cs 25:75%, 50:50%, dan  
75:25%



Larutan *Congo Red* sebelum adsorpsi



Larutan *Congo Red* setelah di adsorpsi  
oleh Zeolit Teraktivasi asam



Larutan *Congo Red* setelah di adsorpsi oleh  
komposit Z-Cs 25:75%



Larutan *Congo Red* setelah di adsorpsi  
oleh komposit Z-Cs 50:50%

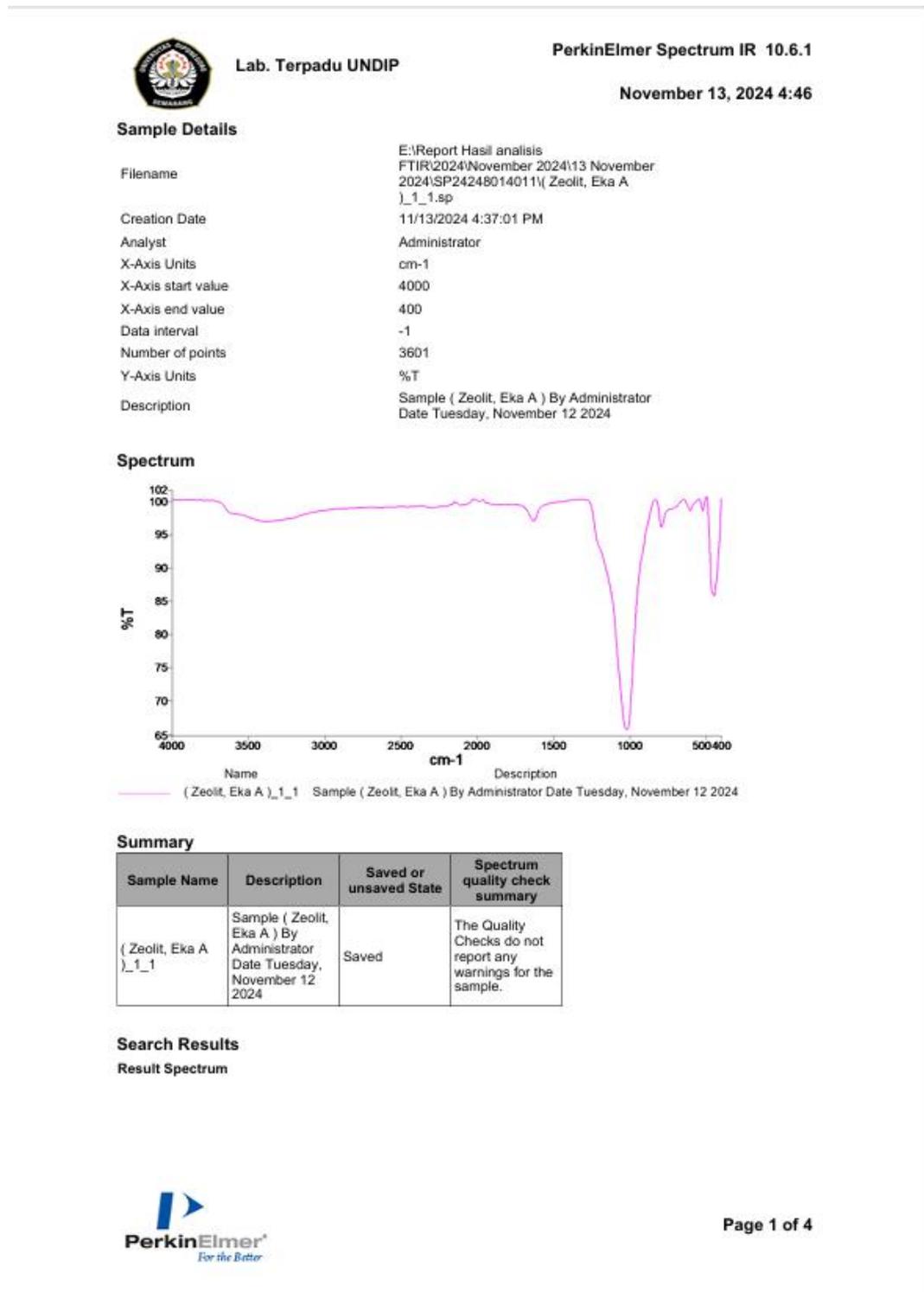


Larutan *Congo Red* setelah di adsorpsi oleh  
komposit Z-Cs 75:25%

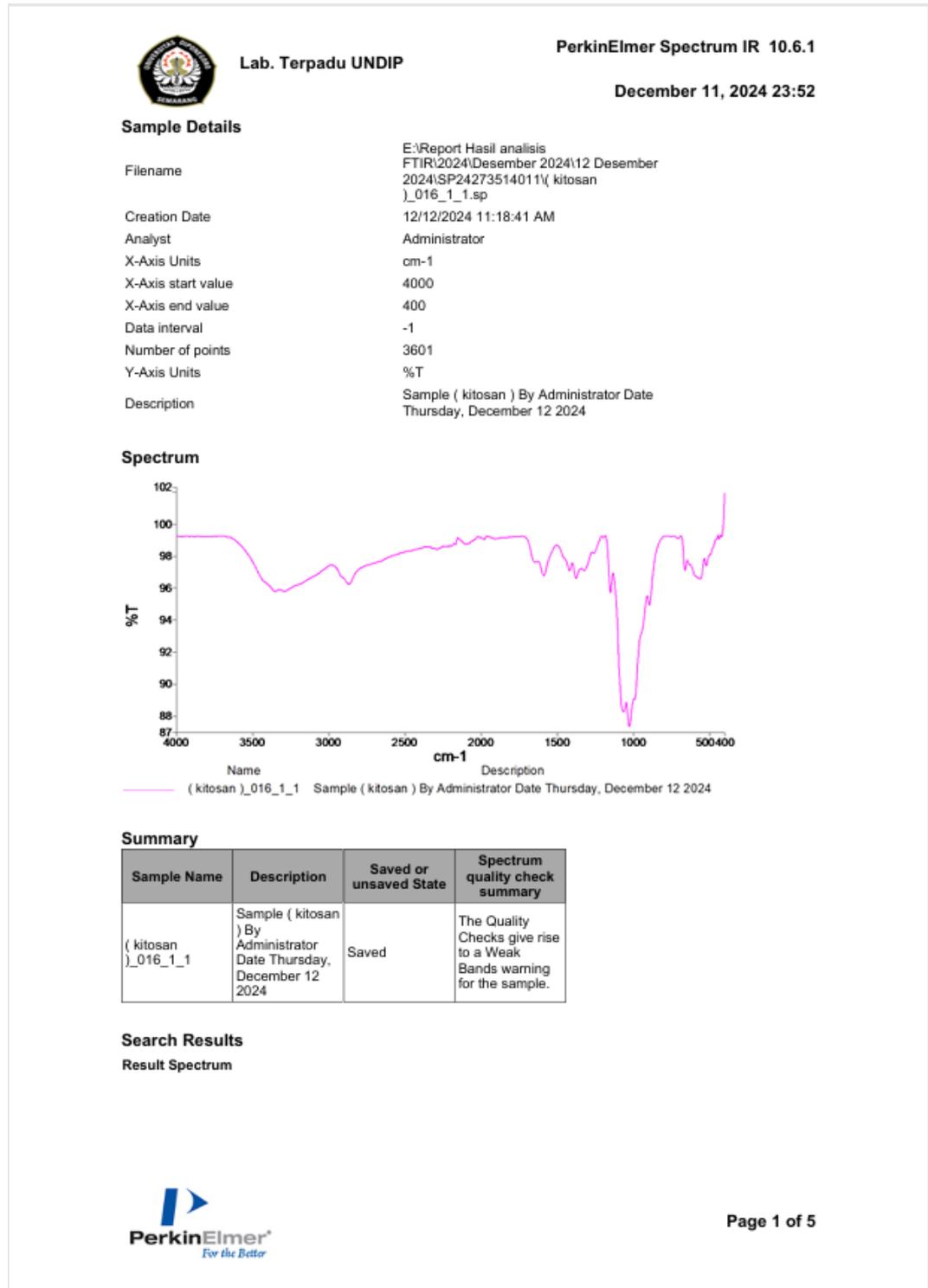
## Lampiran 5. Hasil Analisis dan Karakterisasi

### 5.1 Karakterisasi FTIR

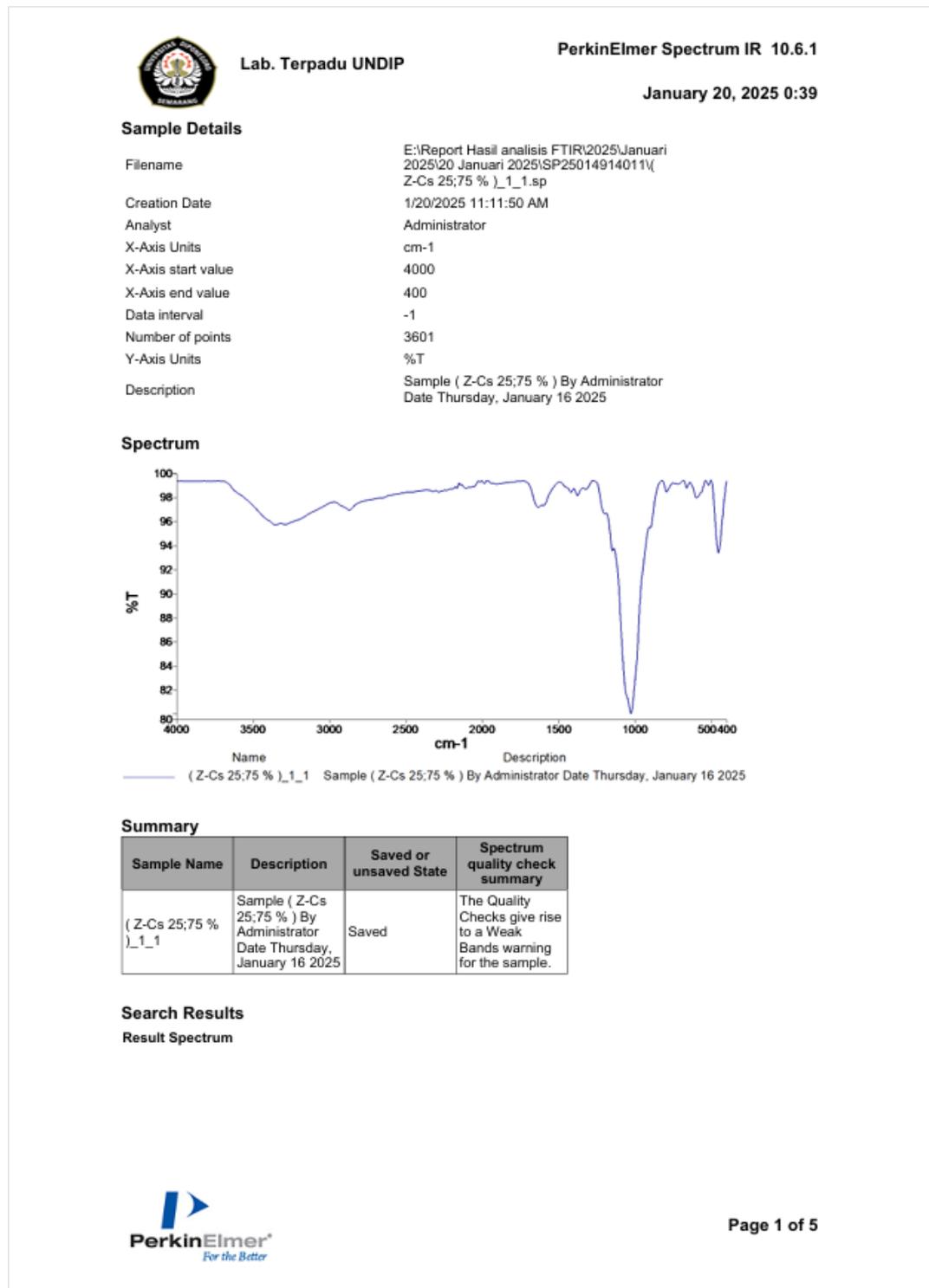
#### a. FTIR Zeolit Alam Aktivasi



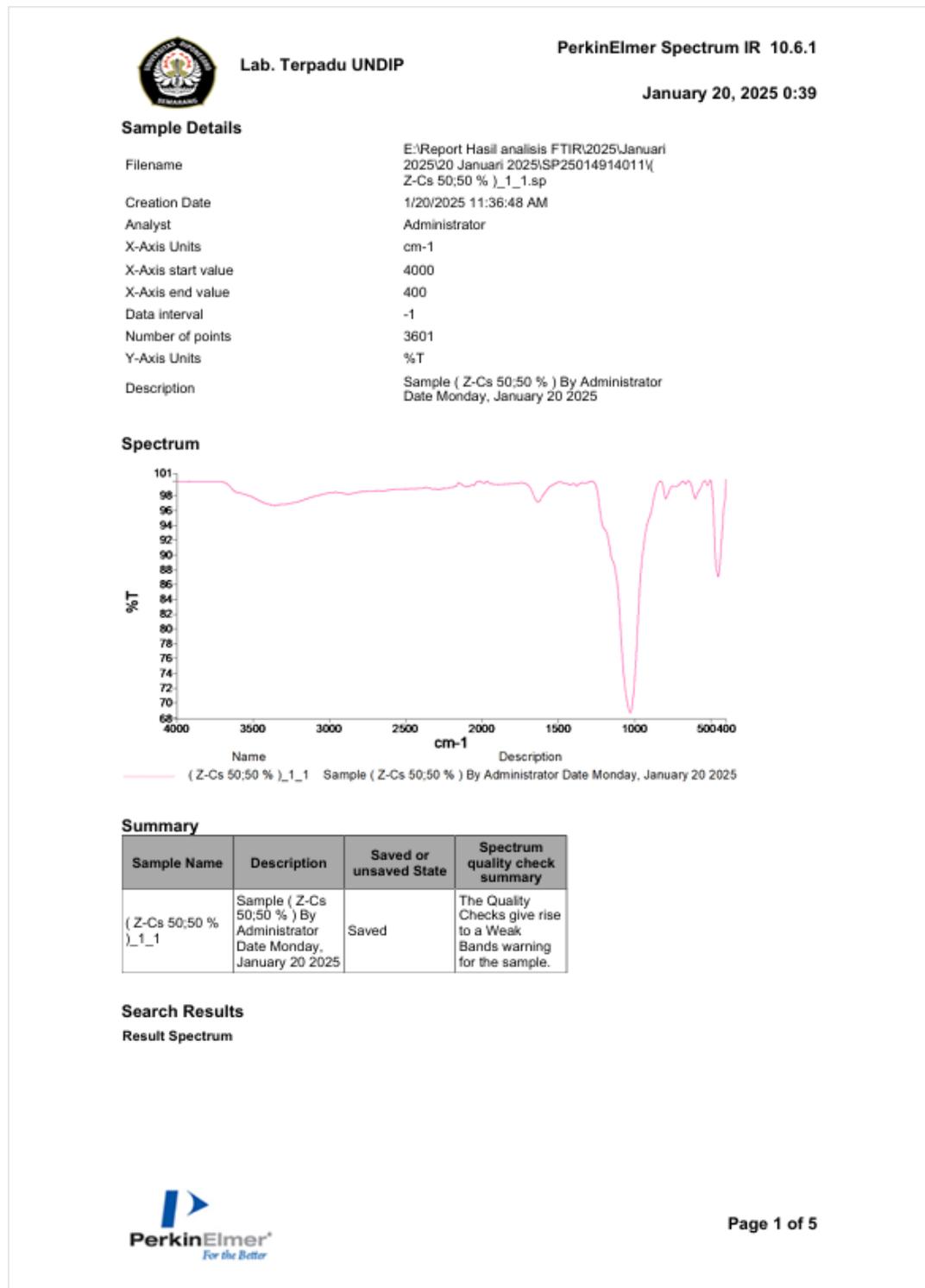
## b. FTIR Kitosan



## c. FTIR Z-Cs 25:75%



## d. FTIR Z-Cs 50:50%



## e. FTIR Z-Cs 75:25%



Lab. Terpadu UNDIP

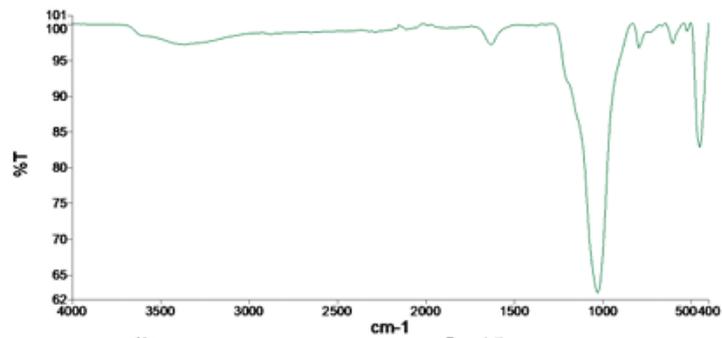
PerkinElmer Spectrum IR 10.6.1

January 20, 2025 0:39

## Sample Details

Filename E:\Report Hasil analisis FTIR\2025\Januari  
 2025\20 Januari 2025\SP25014914011\  
 Z-Cs 75;25 % )\_1\_1.sp  
 Creation Date 1/20/2025 11:40:00 AM  
 Analyst Administrator  
 X-Axis Units cm-1  
 X-Axis start value 4000  
 X-Axis end value 400  
 Data interval -1  
 Number of points 3601  
 Y-Axis Units %T  
 Description Sample ( Z-Cs 75;25 % ) By Administrator  
 Date Monday, January 20 2025

## Spectrum



Name ( Z-Cs 75;25 % )\_1\_1 Description Sample ( Z-Cs 75;25 % ) By Administrator Date Monday, January 20 2025

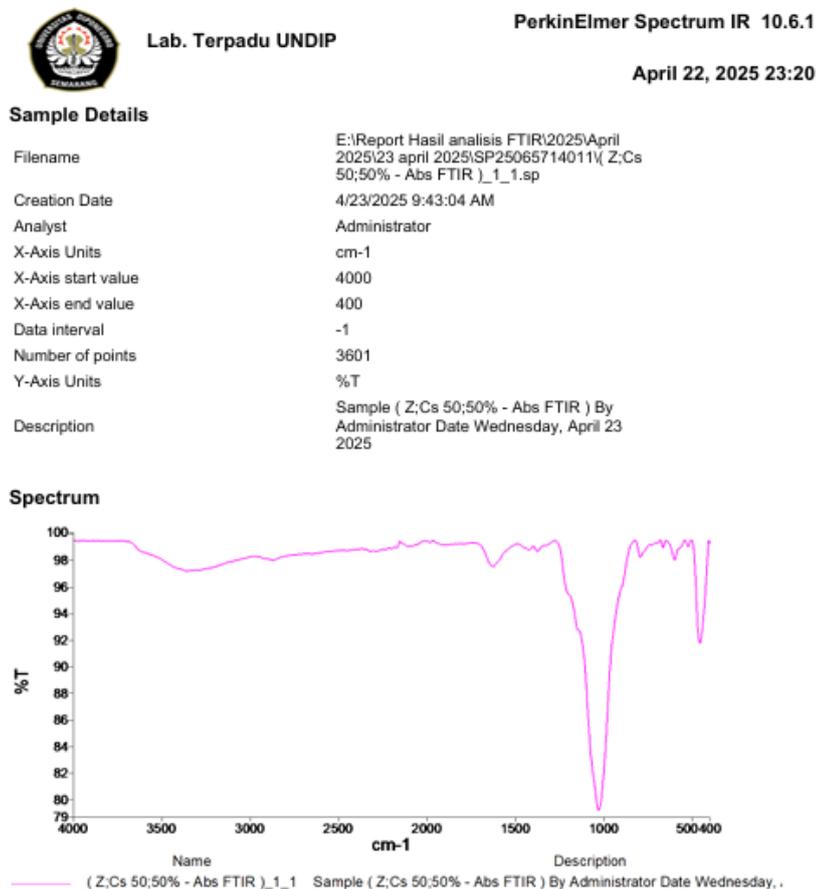
## Summary

Sample Name	Description	Saved or unsaved State	Spectrum quality check summary
( Z-Cs 75;25 % )_1_1	Sample ( Z-Cs 75;25 % ) By Administrator Date Monday, January 20 2025	Saved	The Quality Checks do not report any warnings for the sample.

## Search Results

Result Spectrum

## f. FTIR Z-Cs setelah adsorpsi

**Summary**

Sample Name	Description	Saved or unsaved State	Spectrum quality check summary
( Z;Cs 50;50% - Abs FTIR )_1_1	Sample ( Z;Cs 50;50% - Abs FTIR ) By Administrator Date Wednesday, April 23 2025	Saved	The Quality Checks give rise to a Weak Bands warning for the sample.

**Search Results****Result Spectrum**

## 5. 2 Karakterisasi PSA

### a. Zeolit alam teraktivasi

Bettersize

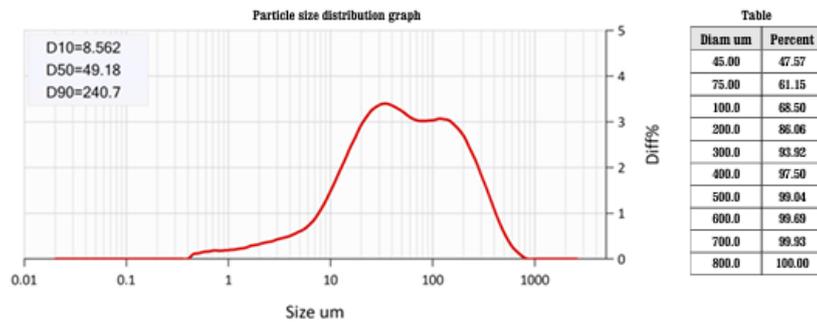
### Bettersizer 2600(Wet) Particle Size Analysis Report

Range : 0.02um - 2600um

Sample : Zeolit Alam	Number : 1	Sample Source : UNZIP Kimia
Operator : Fath	Test Time : 2025-03-10 18:41:09	Measured By : NEXUS Analytics Indonesia
Particle RI : 1.52	Particle AR : 0.05	Medium RI : 1.333
Optical : Mie	Mode : Adaptive	Distribution : Volume
Sampling :	Speed (rpm):	Ultrasound:
Medium : Water	Dispersant : Water	Remark :

D[4,3] : 92.72 um	D[3,2] : 14.53 um	D[1,0] : 0.748 um	Peak Size : 33.47 um
SPAN : 4.720	SSA : 8257m <sup>2</sup> /kg	Residual : 0.343 %	Obscuration : 10.71 %

D03 = 2.240 um	D06 = 4.846 um	D10 = 8.562 um	D16 = 13.21 um	D25 = 20.08 um
D75 = 128.5 um	D50 = 49.18 um	D84 = 183.4 um	D90 = 240.7 um	D97 = 380.1 um



Diam um	Diff%	Cum%	Diam um	Diff%	Cum%	Diam um	Diff%	Cum%	Diam um	Diff%	Cum%
0.020-0.022	0.00	0.00	0.379-0.427	0.01	0.01	7.211-8.112	1.01	9.46	136.9-154.0	3.03	79.72
0.022-0.025	0.00	0.00	0.427-0.480	0.11	0.12	8.112-9.126	1.21	10.67	153.0-173.2	2.93	82.65
0.025-0.028	0.00	0.00	0.480-0.540	0.13	0.25	9.126-10.26	1.43	12.10	173.2-194.9	2.81	85.46
0.028-0.032	0.00	0.00	0.540-0.608	0.16	0.41	10.26-11.54	1.67	13.77	194.9-219.3	2.64	88.10
0.032-0.036	0.00	0.00	0.608-0.684	0.17	0.58	11.54-12.99	1.93	15.70	219.3-246.7	2.40	90.50
0.036-0.040	0.00	0.00	0.684-0.769	0.19	0.77	12.99-14.61	2.17	17.87	246.7-277.5	2.16	92.66
0.040-0.045	0.00	0.00	0.769-0.865	0.18	0.95	14.61-16.44	2.44	20.31	277.5-312.2	1.87	94.53
0.045-0.051	0.00	0.00	0.865-0.974	0.19	1.14	16.44-18.48	2.67	22.98	312.2-351.2	1.58	96.11
0.051-0.057	0.00	0.00	0.974-1.095	0.20	1.34	18.48-20.80	2.91	25.89	351.2-395.1	1.27	97.38
0.057-0.064	0.00	0.00	1.095-1.232	0.21	1.55	20.80-23.40	3.09	28.98	395.1-444.5	0.97	98.35
0.064-0.073	0.00	0.00	1.232-1.386	0.23	1.78	23.40-26.33	3.24	32.22	444.5-500.0	0.69	99.04
0.073-0.082	0.00	0.00	1.386-1.560	0.25	2.03	26.33-29.62	3.33	35.55	500.0-562.5	0.46	99.50
0.082-0.092	0.00	0.00	1.560-1.755	0.29	2.32	29.62-33.32	3.39	38.94	562.5-625.0	0.28	99.78
0.092-0.103	0.00	0.00	1.755-1.974	0.31	2.63	33.32-37.49	3.40	42.34	625.0-690.0	0.16	99.94
0.103-0.116	0.00	0.00	1.974-2.221	0.34	2.97	37.49-42.17	3.36	45.70	690.0-760.0	0.06	100.00
0.116-0.131	0.00	0.00	2.221-2.498	0.37	3.34	42.17-47.44	3.30	49.00	760.0-830.0	0.00	100.00
0.131-0.148	0.00	0.00	2.498-2.811	0.39	3.73	47.44-53.37	3.23	52.23	830.0-900.0	0.00	100.00
0.148-0.166	0.00	0.00	2.811-3.162	0.43	4.16	53.37-60.04	3.14	55.37	900.0-1013	0.00	100.00
0.166-0.187	0.00	0.00	3.162-3.567	0.46	4.62	60.04-67.55	3.07	58.44	1013-1140	0.00	100.00
0.187-0.210	0.00	0.00	3.567-4.002	0.49	5.11	67.55-75.98	3.03	61.47	1140-1282	0.00	100.00
0.210-0.237	0.00	0.00	4.002-4.502	0.53	5.64	75.98-85.49	3.02	64.49	1282-1443	0.00	100.00
0.237-0.266	0.00	0.00	4.502-5.065	0.59	6.23	85.49-96.17	3.03	67.52	1443-1623	0.00	100.00
0.266-0.300	0.00	0.00	5.065-5.698	0.64	6.87	96.17-108.1	3.04	70.56	1623-1826	0.00	100.00
0.300-0.337	0.00	0.00	5.698-6.410	0.73	7.60	108.1-121.7	3.07	73.63	1826-2054	0.00	100.00
0.337-0.379	0.00	0.00	6.410-7.211	0.85	8.45	121.7-136.9	3.06	76.69	2054-2311	0.00	100.00

Bettersize Instruments Ltd. www.bettersizeinstruments.com Email:services@bettersize.com Tel:+86 (415) 618 4440

System Status: A:0-0-0-0-0-0-10-0-25-0-75-0-90-1-1 E:87-0-0-3 J:14-0-10-0-5 K:1-12-1-1-0-1 L:1-50-1-20000-0-1 M:0-3-1-1-0-7-5-1-1-2-1-23-1-30-10-0-25-0-0-0-0  
 SW:01.D1.00204.00 01.10.BE.002540 PWD:Time:2025-03-11 10:10:52 Pkg:Name: Pak-Parlayo UNZIP Kimia Record No.: 7 PWT: Bettersizer 2600 BT-BSN Pro General Test Method: Laser

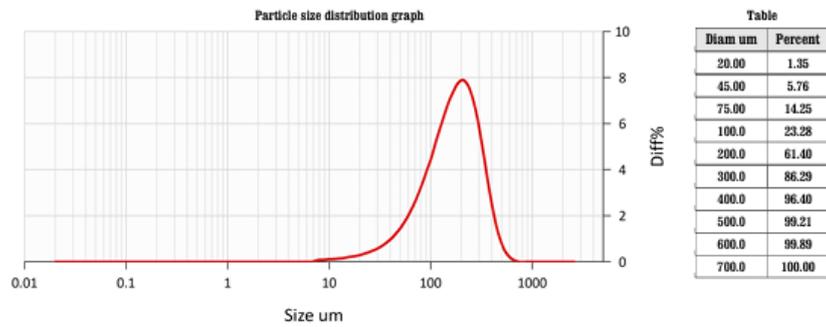
b. Kitosan



Betztersizer 2600(Wet) Particle Size Analysis Report

Range : 0.02um - 2600um

<b>Sample :</b> Kitosan	<b>Number :</b> 1	<b>Sample Source :</b> UNDIP	
<b>Operator :</b> Fath	<b>Test Time :</b> 2025-04-25 14:30:44	<b>Measured By :</b> NEXUS Analytics Indonesia	
<b>Particle RI :</b> 1.5	<b>Particle AR :</b> 0.1	<b>Medium RI :</b> 1.333	
<b>Optical :</b> Mie	<b>Mode :</b> Adaptive	<b>Distribution :</b> Volume	
<b>Sampling:</b>	<b>Speed (rpm):</b>	<b>Ultrasound:</b>	
<b>Medium :</b> Water	<b>Dispersant :</b> Water	<b>Remark :</b>	
<b>D[4,3] :</b> 182.9 um	<b>D[3,2] :</b> 111.5 um	<b>D[1,0] :</b> 18.46 um	<b>Peak Size :</b> 205.1 um
<b>SPAN :</b> 1.560	<b>SSA :</b> 19.91m <sup>2</sup> /kg	<b>Residual :</b> 0.537 %	<b>Obscuration :</b> 11.92 %
<b>D03 =</b> 31.17 um	<b>D06 =</b> 46.01 um	<b>D10 =</b> 61.28 um	<b>D16 =</b> 80.11 um
<b>D75 =</b> 245.8 um	<b>D50 =</b> 168.1 um	<b>D84 =</b> 285.6 um	<b>D90 =</b> 323.5 um
<b>D25 =</b> 104.4 um		<b>D97 =</b> 410.0 um	



Diam um	Diff%	Cum%	Diam um	Diff%	Cum%	Diam um	Diff%	Cum%	Diam um	Diff%	Cum%
0.020-0.022	0.00	0.00	0.379-0.427	0.00	0.00	7.211-8.112	0.08	0.09	136.9-154.0	6.78	44.61
0.022-0.025	0.00	0.00	0.427-0.480	0.00	0.00	8.112-9.126	0.09	0.18	154.0-173.2	7.32	51.93
0.025-0.028	0.00	0.00	0.480-0.540	0.00	0.00	9.126-10.26	0.11	0.29	173.2-194.9	7.74	59.67
0.028-0.032	0.00	0.00	0.540-0.608	0.00	0.00	10.26-11.54	0.12	0.41	194.9-219.3	7.90	67.57
0.032-0.036	0.00	0.00	0.608-0.684	0.00	0.00	11.54-12.99	0.14	0.55	219.3-246.7	7.65	75.22
0.036-0.040	0.00	0.00	0.684-0.769	0.00	0.00	12.99-14.61	0.16	0.71	246.7-277.5	7.02	82.24
0.040-0.045	0.00	0.00	0.769-0.865	0.00	0.00	14.61-16.44	0.21	0.92	277.5-312.2	5.98	88.22
0.045-0.051	0.00	0.00	0.865-0.974	0.00	0.00	16.44-18.48	0.24	1.16	312.2-351.2	4.65	92.87
0.051-0.057	0.00	0.00	0.974-1.095	0.00	0.00	18.48-20.80	0.29	1.45	351.2-395.1	3.24	96.11
0.057-0.064	0.00	0.00	1.095-1.232	0.00	0.00	20.80-23.40	0.35	1.80	395.1-444.5	2.90	98.11
0.064-0.073	0.00	0.00	1.232-1.386	0.00	0.00	23.40-26.33	0.43	2.23	444.5-500.0	1.10	99.21
0.073-0.082	0.00	0.00	1.386-1.560	0.00	0.00	26.33-29.62	0.52	2.75	500.0-562.5	0.52	99.73
0.082-0.092	0.00	0.00	1.560-1.755	0.00	0.00	29.62-33.32	0.63	3.38	562.5-632.8	0.21	99.94
0.092-0.103	0.00	0.00	1.755-1.974	0.00	0.00	33.32-37.49	0.79	4.17	632.8-711.9	0.06	100.00
0.103-0.116	0.00	0.00	1.974-2.221	0.00	0.00	37.49-42.17	0.97	5.14	711.9-800.9	0.00	100.00
0.116-0.131	0.00	0.00	2.221-2.498	0.00	0.00	42.17-47.44	1.20	6.34	800.9-900.9	0.00	100.00
0.131-0.148	0.00	0.00	2.498-2.811	0.00	0.00	47.44-53.37	1.48	7.82	900.9-1013	0.00	100.00
0.148-0.166	0.00	0.00	2.811-3.162	0.00	0.00	53.37-60.04	1.82	9.64	1013-1140	0.00	100.00
0.166-0.187	0.00	0.00	3.162-3.557	0.00	0.00	60.04-67.55	2.23	11.87	1140-1282	0.00	100.00
0.187-0.210	0.00	0.00	3.557-4.002	0.00	0.00	67.55-75.98	2.72	14.59	1282-1443	0.00	100.00
0.210-0.237	0.00	0.00	4.002-4.502	0.00	0.00	75.98-85.49	3.29	17.88	1443-1623	0.00	100.00
0.237-0.266	0.00	0.00	4.502-5.065	0.00	0.00	85.49-96.17	3.93	21.81	1623-1826	0.00	100.00
0.266-0.300	0.00	0.00	5.065-5.698	0.00	0.00	96.17-108.1	4.58	26.39	1826-2054	0.00	100.00
0.300-0.337	0.00	0.00	5.698-6.410	0.00	0.00	108.1-121.7	5.37	31.76	2054-2311	0.00	100.00
0.337-0.379	0.00	0.00	6.410-7.211	0.01	0.01	121.7-136.9	6.07	37.83	2311-2600	0.00	100.00



d. Z-Cs 50:50%



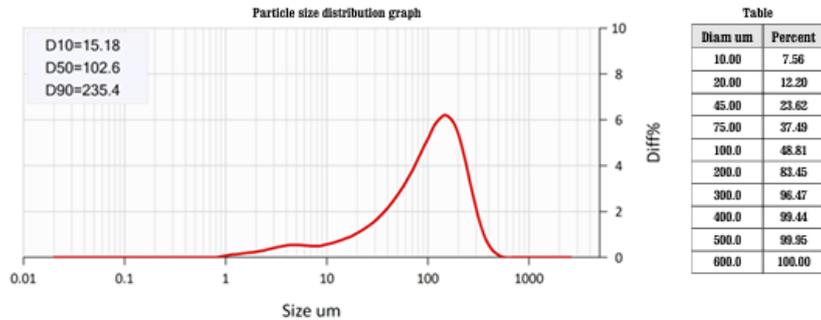
### Betterson 2600(Wet) Particle Size Analysis Report

Range : 0.02um - 2600um

Sample : Zeolit Alam - Kitosan 50\_50 Number : 3 Sample Source : UNZIP Kimia  
 Operator : Fath Test Time : 2025-03-10 19:11:31 Measured By : NEXUS Analytics Indonesia  
 Particle RI : 1.5 Particle AR : 0.05 Medium RI : 1.333  
 Optical : Mie Mode : Adaptive Distribution : Volume  
 Sampling : Speed (rpm): Ultrasound:  
 Medium : Water Dispersant : Water Remark :

D[4.3] : 116.5 um D[3.2] : 27.45 um D[1.0] : 1.725 um Peak Size : 149.9 um  
 SPAN : 2.144 SSA : 4370m<sup>2</sup>/kg Residual : 0.450 % Obscuration : 10.44 %

D03 = 3.644 um D06 = 7.038 um D10 = 15.18 um D16 = 28.38 um D25 = 47.97 um  
 D75 = 168.6 um D50 = 102.6 um D84 = 202.6 um D90 = 235.4 um D97 = 309.4 um



Diam um	Diff%	Cum%	Diam um	Diff%	Cum%	Diam um	Diff%	Cum%	Diam um	Diff%	Cum%
0.020-0.022	0.00	0.00	0.379-0.427	0.00	0.00	7.211-8.112	6.50	6.51	135.9-154.0	6.21	70.35
0.022-0.025	0.00	0.00	0.427-0.450	0.00	0.00	8.112-9.126	0.51	7.12	154.0-173.2	6.10	76.45
0.025-0.028	0.00	0.00	0.450-0.500	0.00	0.00	9.126-10.26	0.57	7.69	173.2-194.9	5.79	82.24
0.028-0.032	0.00	0.00	0.500-0.608	0.00	0.00	10.26-11.54	0.61	8.30	194.9-219.3	5.15	87.39
0.032-0.036	0.00	0.00	0.608-0.684	0.00	0.00	11.54-12.99	0.68	8.98	219.3-246.7	4.26	91.65
0.036-0.040	0.00	0.00	0.684-0.769	0.00	0.00	12.99-14.61	0.76	9.74	246.7-277.5	3.24	94.89
0.040-0.045	0.00	0.00	0.769-0.865	0.01	0.01	14.61-16.44	0.94	10.58	277.5-312.2	2.26	97.15
0.045-0.051	0.00	0.00	0.865-0.974	0.06	0.07	16.44-18.48	0.90	11.51	312.2-351.2	1.42	98.57
0.051-0.057	0.00	0.00	0.974-1.095	0.10	0.17	18.48-20.80	1.05	12.56	351.2-395.1	0.80	99.37
0.057-0.064	0.00	0.00	1.095-1.232	0.13	0.30	20.80-23.40	1.18	13.74	395.1-444.5	0.40	99.77
0.064-0.073	0.00	0.00	1.232-1.386	0.15	0.45	23.40-26.33	1.33	15.07	444.5-500.0	0.18	99.95
0.073-0.082	0.00	0.00	1.386-1.560	0.18	0.63	26.33-29.62	1.49	16.56	500.0-562.5	0.05	100.00
0.082-0.092	0.00	0.00	1.560-1.755	0.21	0.84	29.62-33.32	1.68	18.24	562.5-632.8	0.00	100.00
0.092-0.103	0.00	0.00	1.755-1.974	0.23	1.07	33.32-37.49	1.91	20.15	632.8-711.9	0.00	100.00
0.103-0.116	0.00	0.00	1.974-2.221	0.27	1.34	37.49-42.17	2.16	22.31	711.9-800.9	0.00	100.00
0.116-0.131	0.00	0.00	2.221-2.498	0.31	1.65	42.17-47.44	2.44	24.75	800.9-900.9	0.00	100.00
0.131-0.148	0.00	0.00	2.498-2.811	0.36	2.01	47.44-53.37	2.75	27.50	900.9-1013	0.00	100.00
0.148-0.166	0.00	0.00	2.811-3.162	0.42	2.43	53.37-60.04	3.09	30.59	1013-1140	0.00	100.00
0.166-0.187	0.00	0.00	3.162-3.557	0.47	2.90	60.04-67.55	3.47	34.06	1140-1282	0.00	100.00
0.187-0.210	0.00	0.00	3.557-4.002	0.51	3.41	67.55-75.98	3.88	37.94	1282-1443	0.00	100.00
0.210-0.237	0.00	0.00	4.002-4.502	0.55	3.96	75.98-85.49	4.35	42.29	1443-1623	0.00	100.00
0.237-0.266	0.00	0.00	4.502-5.065	0.56	4.52	85.49-96.17	4.82	47.11	1623-1826	0.00	100.00
0.266-0.300	0.00	0.00	5.065-5.698	0.55	5.07	96.17-108.1	5.26	52.37	1826-2054	0.00	100.00
0.300-0.337	0.00	0.00	5.698-6.410	0.53	5.60	108.1-121.7	5.74	58.11	2054-2311	0.00	100.00
0.337-0.379	0.00	0.00	6.410-7.211	0.51	6.11	121.7-136.9	6.03	64.14	2311-2600	0.00	100.00

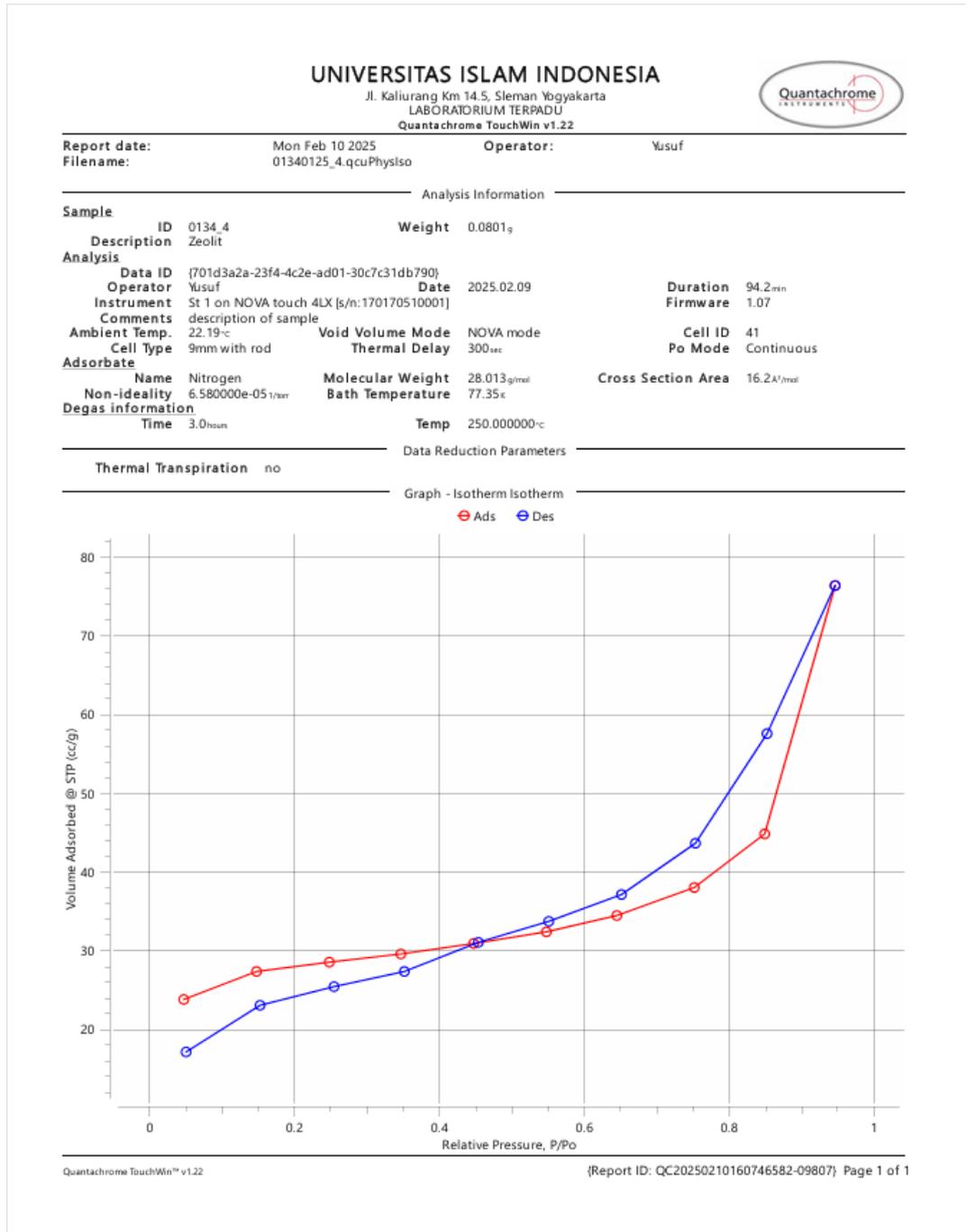
Betterson Instruments Ltd. www.bettersoninstruments.com Email:services@betterson.com Tel: +86 (415) 618 4440

System Status: A-0-0-0-0-0-0-0-10-0-25-0-75-0-90-1-1 E-87-0-0-6-3 J-1-4-0-10-0-5 K-1-12-1-1-0-1 L-1-50-1-20000-0-1 M-0-3-1-1-0-7-5-1-1-2-1-0-3-1-30-10-0-25-0-0-0-5  
 SW:01.01.08204.00 01.10.08.002540 Print Time: 2025-03-11 15:13:39 Project Name: Pak Parlayo UNZIP Kimia Record No.: 17 PMT: Betterson 2600 BT-85N Pro General Test Method: Laser



### 5.3 Karakterisasi SAA

#### a. Zeolit alam teraktivasi



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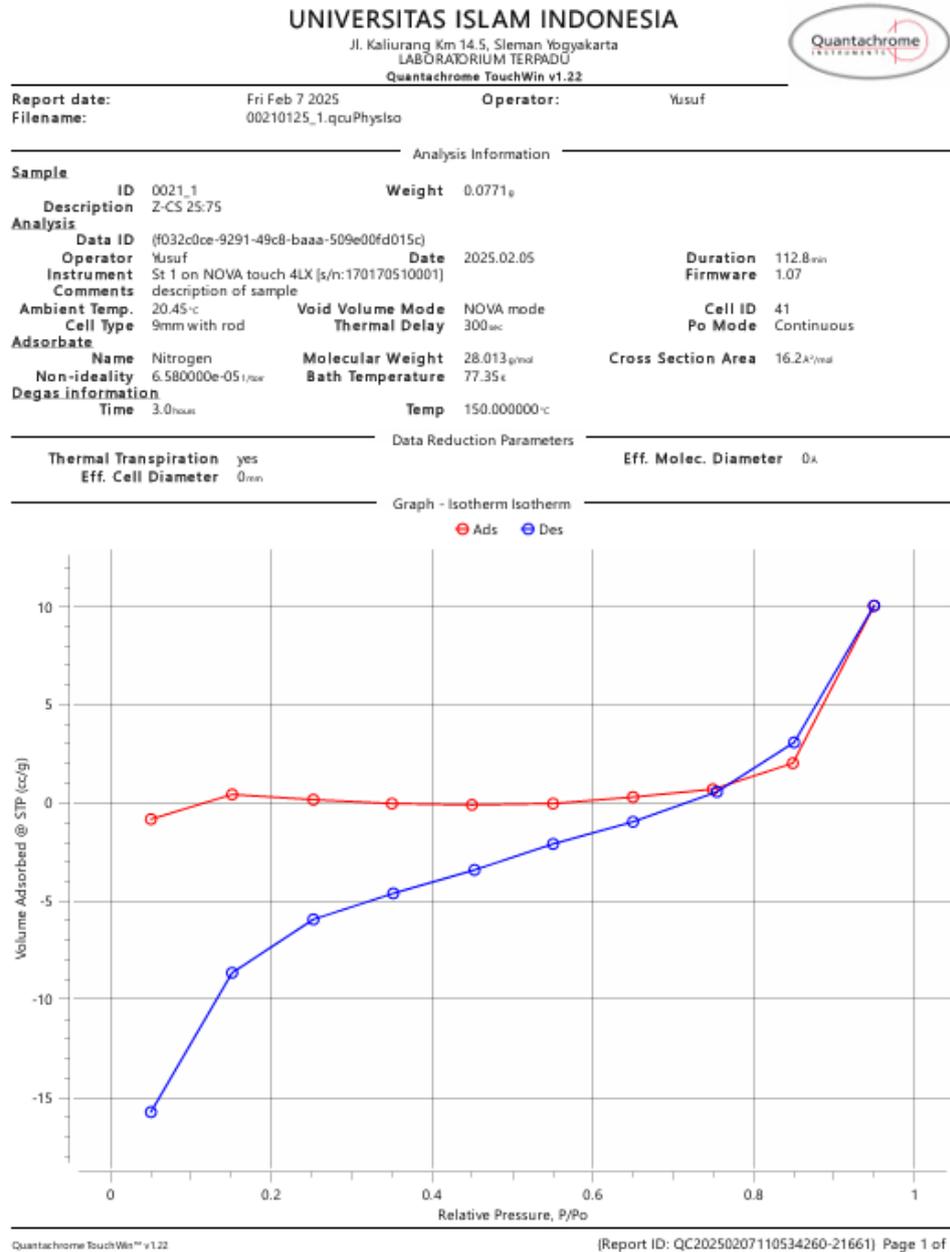
Jl. Kaliurang Km 14.5, Sleman Yogyakarta  
LABORATORIUM TERPADU  
Quantachrome TouchWin v1.22



Report date: Mon Feb 10 2025 Operator: Yusuf  
Filename: 01340125\_4.qcuPhysIso

Sample		Analysis Information	
ID	0134_4	Weight	0.0801 <sub>g</sub>
Description	Zeolit		
<b>Analysis</b>			
Data ID	{701d3a2a-23f4-4c2e-ad01-30c7c31db790}	Date	2025.02.09
Operator	Yusuf	Duration	94.2 <sub>min</sub>
Instrument	St 1 on NOVA touch 4LX [s/n:170170510001]	Firmware	1.07
Comments	description of sample		
Ambient Temp.	22.19 <sub>°c</sub>	Void Volume Mode	NOVA mode
Cell Type	9mm with rod	Thermal Delay	300 <sub>sec</sub>
Adsorbate		Cell ID	41
Name	Nitrogen	Molecular Weight	28.013 <sub>g/mol</sub>
Non-ideality	6.580000e-05 <sub>1/ton</sub>	Bath Temperature	77.35 <sub>K</sub>
Degas information		Cross Section Area	16.2 <sub>A<sup>2</sup>/mol</sub>
Time	3.0 <sub>hours</sub>	Temp	250.000000 <sub>°c</sub>
<b>Data Reduction Parameters</b>			
Thermal Transpiration	no		
Temp. Comp	no		
Thickness Method	deBoer		
P-tags below 0.35	ignored	Moving Pt. Average	off
Adsorbate Model		Cross Section Area	16.2 <sub>A<sup>2</sup>/molec</sub>
Name	Nitrogen	Molecular Weight	28.0134 <sub>g</sub>
Bath Temperature	77.35 <sub>K</sub>		
<b>Area-Volume Summary results</b>			
<b>Surface Area Results</b>			
Multipoint BET	84.0021 <sub>m<sup>2</sup>/g</sub>		
BJH adsorption	21.1121 <sub>m<sup>2</sup>/g</sub>		
BJH desorption	31.8111 <sub>m<sup>2</sup>/g</sub>		
<b>Pore Volume Results</b>			
BJH adsorption cumulative micropore volume	0.0825801 <sub>cc/g</sub>		
BJH desorption cumulative micropore volume	0.0882382 <sub>cc/g</sub>		
Total Pore Volume	0.118494 <sub>cc/g</sub>		
<b>Pore Size Results</b>			
BJH adsorption pore radius	12.7481 <sub>nm</sub>		
BJH desorption pore radius	5.74617 <sub>nm</sub>		
Average Pore Size	2.82122 <sub>nm</sub>		

b. Z-Cs 25:75%


Quantachrome TouchWin™ v1.22
(Report ID: QC20250207110534260-21661) Page 1 of 1

## UNIVERSITAS ISLAM INDONESIA

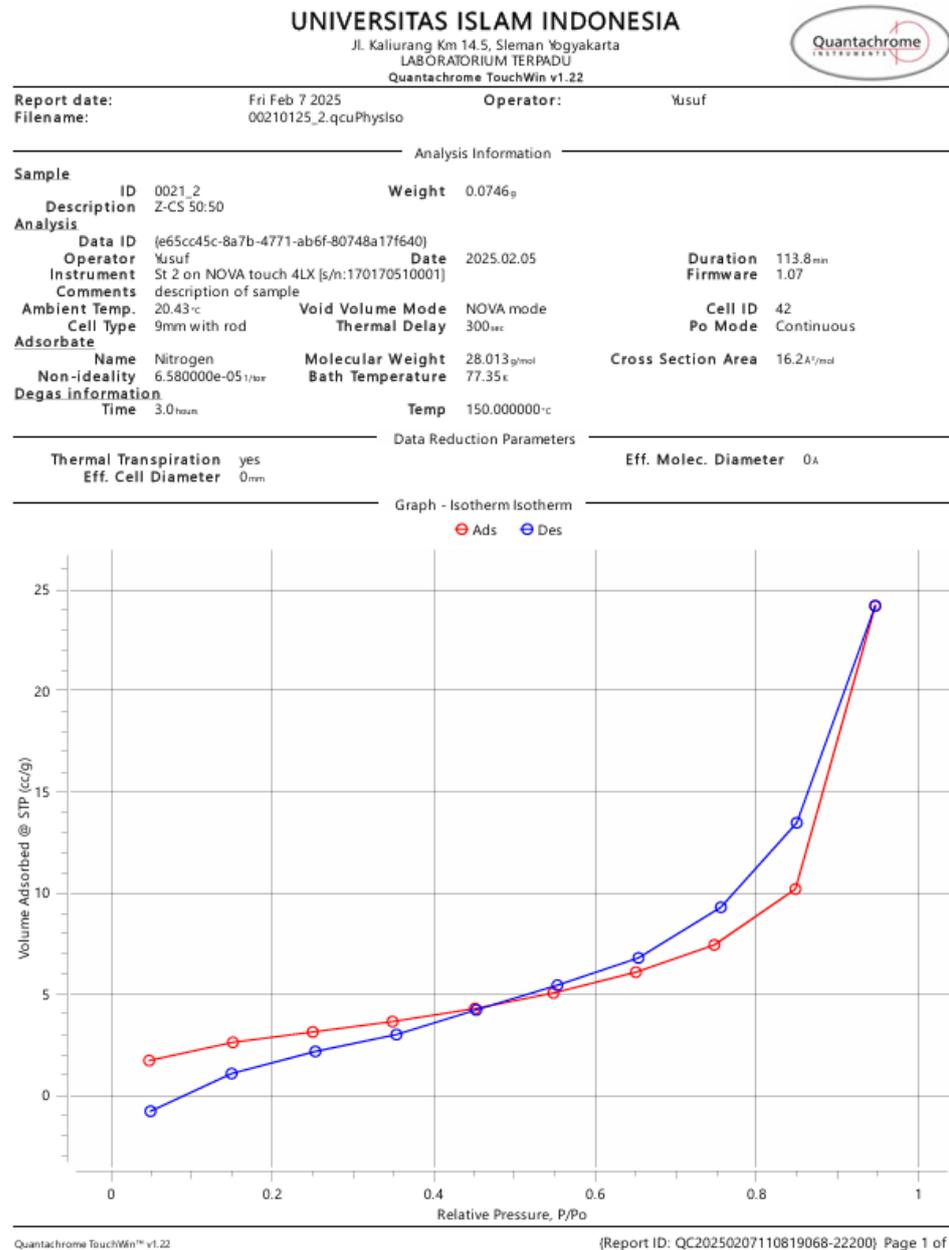
Jl. Kaliurang Km 14.5, Sleman Yogyakarta  
 LABORATORIUM TERPADU  
 Quantachrome TouchWin v1.22



Report date: Fri Feb 7 2025 Operator: Yusuf  
 Filename: 00210125\_1.qcuPhysIso

Sample		Analysis Information	
ID	0021_1	Weight	0.0771g
Description	Z-CS 25:75		
<b>Analysis</b>			
Data ID	{f032c0ce-9291-49c8-baaa-509e00fd015c}	Date	2025.02.05
Operator	Yusuf	Duration	112.8min
Instrument	St 1 on NOVA touch 4LX [s/n:170170510001]	Firmware	1.07
Comments	description of sample		
Ambient Temp.	20.45°C	Void Volume Mode	NOVA mode
Cell Type	9mm with rod	Thermal Delay	300sec
Adsorbate		Cell ID	41
Name	Nitrogen	Molecular Weight	28.013g/mol
Non-ideality	6.580000e-05 1/str	Bath Temperature	77.35K
Degas information		Cross Section Area	16.2A <sup>2</sup> /mol
Time	3.0hours	Temp	150.000000°C
<b>Thermal Transpiration</b>		<b>Data Reduction Parameters</b>	
Temp. Comp	no		
Thickness Method	deBoer		
P-tags below 0.35	ignored		
Adsorbate Model		Moving Pt. Average	off
Name	Nitrogen	Molecular Weight	28.0134g
Bath Temperature	77.35K	Cross Section Area	16.2A <sup>2</sup> /mol
<b>Area-Volume Summary results</b>			
<b>Surface Area Results</b>			
Multipoint BET	0.407347 m <sup>2</sup> /g		
BJH adsorption	3.48024 m <sup>2</sup> /g		
BJH desorption	10.108 m <sup>2</sup> /g		
<b>Pore Volume Results</b>			
BJH adsorption cumulative micropore volume	0.0178836 cc/g		
BJH desorption cumulative micropore volume	0.0264484 cc/g		
Total Pore Volume	0.0155967 cc/g		
<b>Pore Size Results</b>			
BJH adsorption pore radius	13.5055nm		
BJH desorption pore radius	2.0556nm		
Average Pore Size	76.577nm		

## c. Z-Cs 50:50%



## UNIVERSITAS ISLAM INDONESIA

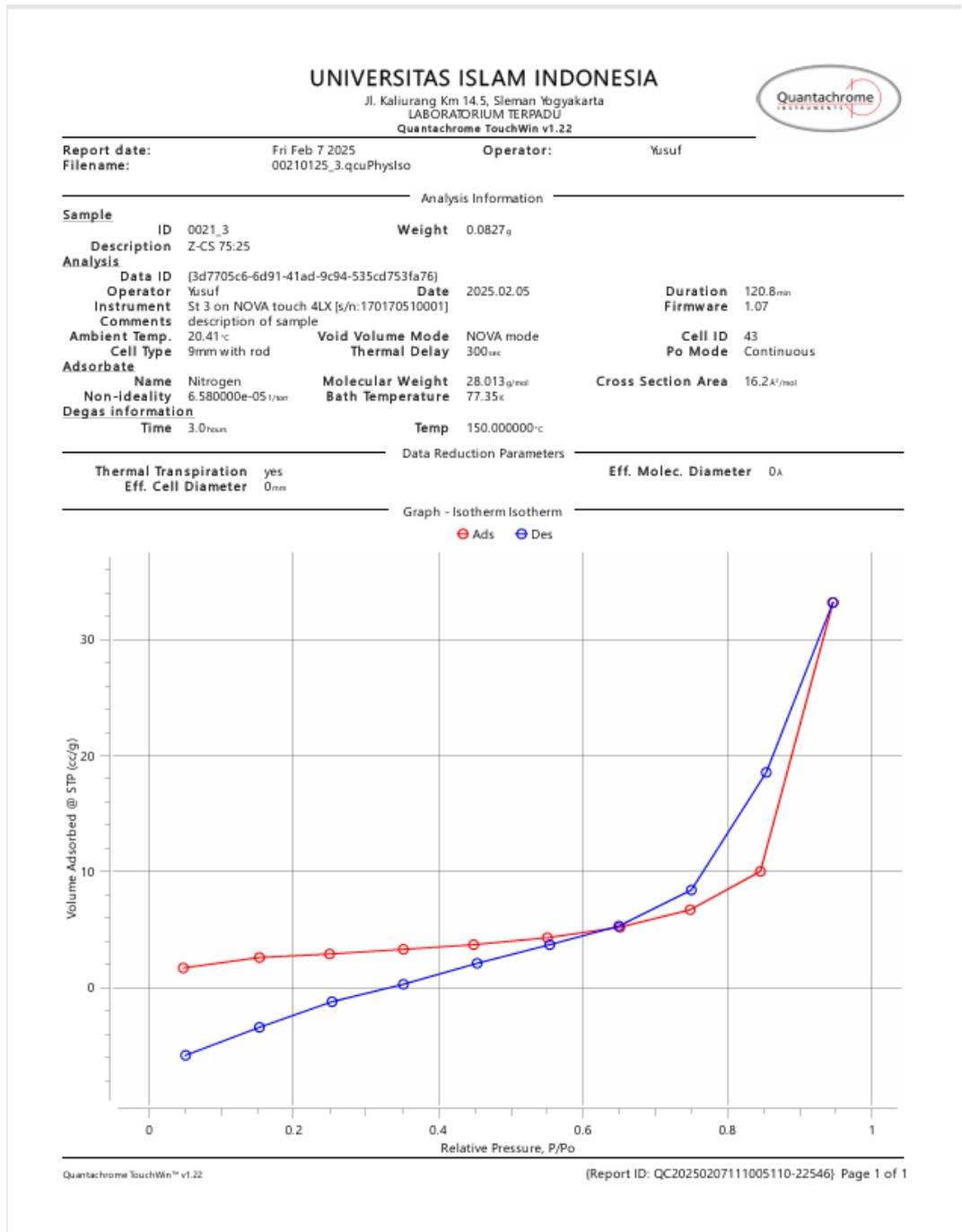
Jl. Kaliurang Km 14.5, Sleman Yogyakarta  
 LABORATORIUM TERPADU  
 Quantachrome TouchWin v1.22



Report date: Fri Feb 7 2025 Operator: Yusuf  
 Filename: 00210125\_2.qcuPhysIso

Sample		Analysis Information	
ID	0021_2	Weight	0.0746g
Description	Z-CS 50:50		
<b>Analysis</b>			
Data ID	[e65cc45c-8a7b-4771-ab6f-80748a17f640]	Date	2025.02.05
Operator	Yusuf	Duration	113.8min
Instrument	St 2 on NOVA touch 4LX [s/n:170170510001]	Firmware	1.07
Comments	description of sample		
Ambient Temp.	20.43 °C	Void Volume Mode	NOVA mode
Cell Type	9mm with rod	Thermal Delay	300 sec
Cell ID	42	Po Mode	Continuous
<b>Adsorbate</b>			
Name	Nitrogen	Molecular Weight	28.013g/mol
Non-ideality	6.580000e-05 1/loop	Bath Temperature	77.35 K
<b>Degas Information</b>			
Time	3.0hour	Temp	150.000000 °C
<b>Data Reduction Parameters</b>			
Thermal Transpiration	no		
Temp. Comp	no		
Thickness Method	deBoer		
P-tags below 0.35	ignored	Moving Pt. Average	off
<b>Adsorbate Model</b>			
Name	Nitrogen	Molecular Weight	28.0134g
Bath Temperature	77.35 K	Cross Section Area	16.2 A <sup>2</sup> /molcc
<b>Area-Volume Summary results</b>			
<b>Surface Area Results</b>			
Multipoint BET	10.8412 m <sup>2</sup> /g		
BJH adsorption	9.47494 m <sup>2</sup> /g		
BJH desorption	12.7616 m <sup>2</sup> /g		
<b>Pore Volume Results</b>			
BJH adsorption cumulative micropore volume	0.0362483 cc/g		
BJH desorption cumulative micropore volume	0.0381591 cc/g		
Total Pore Volume	0.0375393 cc/g		
<b>Pore Size Results</b>			
BJH adsorption pore radius	2.0446 nm		
BJH desorption pore radius	2.06148 nm		
Average Pore Size	6.9253 nm		

d. Z-Cs 75:25%



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LABORATORIUM TERPADU

Quantachrome TouchWin v1.22

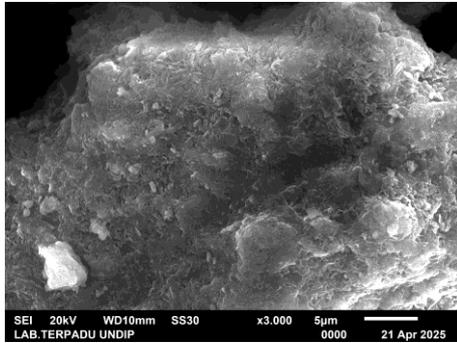


Report date: Fri Feb 7 2025 Operator: Yusuf  
Filename: 00210125\_3.qcuPhysIso

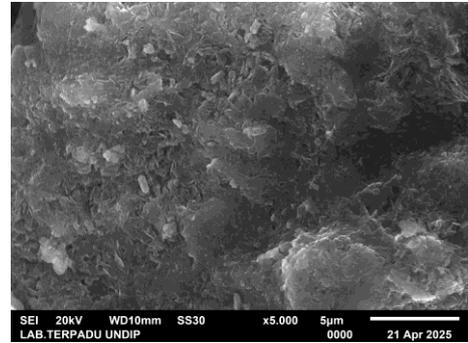
Sample		Analysis Information	
ID	0021_3	Weight	0.0827g
Description	Z-CS 75:25		
<b>Analysis</b>			
Data ID	{3d7705c6-6d91-41ad-9c94-535cd753fa76}	Date	2025.02.05
Operator	Yusuf	Duration	120.8min
Instrument	St 3 on NOVA touch 4LX (s/n:170170510001)	Firmware	1.07
Comments	description of sample		
Ambient Temp.	20.41 °C	Void Volume Mode	NOVA mode
Cell Type	9mm with rod	Thermal Delay	300sec
Adsorbate		Cell ID	43
Name	Nitrogen	Po Mode	Continuous
Molecular Weight	28.013 g/mol	Cross Section Area	16.2 A <sup>2</sup> /mol
Non-ideality	6.580000e-05 1/scr	Bath Temperature	77.35K
<b>Degas information</b>			
Time	3.0hours	Temp	150.000000 °C
<b>Data Reduction Parameters</b>			
Thermal Transpiration	no		
Temp. Comp	no		
Thickness Method	deBoer		
P-tags below 0.35	ignored	Moving Pt. Average	off
Adsorbate Model			
Name	Nitrogen	Molecular Weight	28.0134g
Bath Temperature	77.35K	Cross Section Area	16.2 A <sup>2</sup> /mol
<b>Area-Volume Summary results</b>			
<b>Surface Area Results</b>			
Multipoint BET	9.64066 m <sup>2</sup> /g		
BJH adsorption	10.7226 m <sup>2</sup> /g		
BJH desorption	18.8879 m <sup>2</sup> /g		
<b>Pore Volume Results</b>			
BJH adsorption cumulative micropore volume	0.0519154 cc/g		
BJH desorption cumulative micropore volume	0.0589352 cc/g		
Total Pore Volume	0.051458 cc/g		
<b>Pore Size Results</b>			
BJH adsorption pore radius	12.6792 nm		
BJH desorption pore radius	5.73384 nm		
Average Pore Size	10.6752 nm		

## 5.4 Karakterisasi SEM

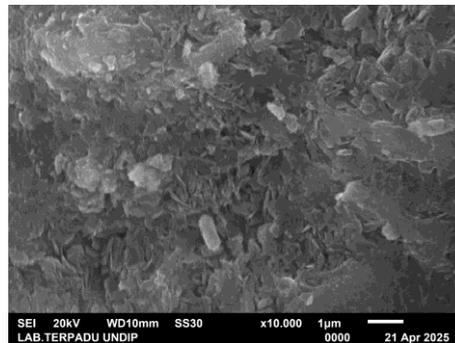
### a. Zeolit alam teraktivasi



a)



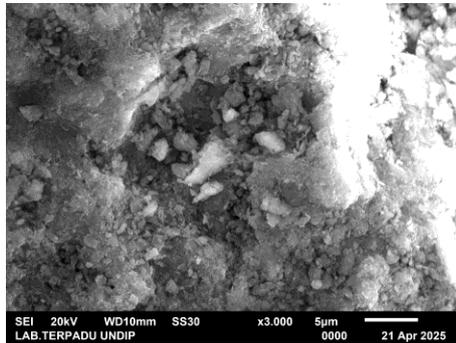
b)



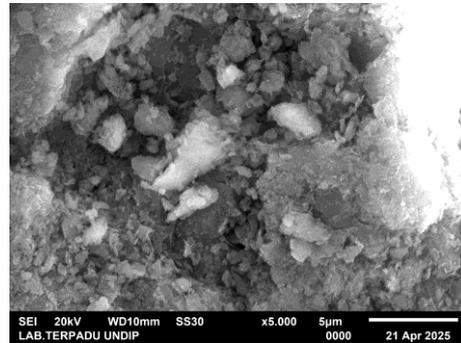
c)

**Gambar L.1** Morfologi permukaan zeolit alam teraktivasi perbesaran (a)3000x, (b) 5000x, dan (c) 10000x

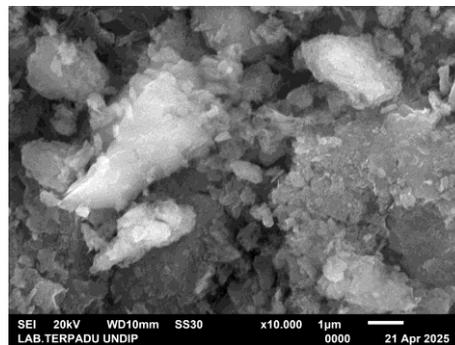
## b. Komposit Z-Cs 50:50%



a)



b)



c)

**Gambar L.2** Morfologi permukaan Z-Cs 50:50% perbesaran a) 3000x, b) 5000x, dan c) 10000x

5.5 Karakterisasi XRD zeolit alam Klaten



**HASIL ANALISIS CONTOH TANAH** DF.7.8.2. b

Nomor SPK : CE 1/07 21/140  
 Nama Pemohon : Aldian Fahrialam  
 Alamat Pemohon : Jl. Sacewa No. 10A, RT 52 / RW 11 Wirebrajan, Yogyakarta  
 Asal Sampel : Tegalmrejo, Gedangsari, Gunungkidul, DIY  
 Uraian Kondisi Sampel Uji : Utuh  
 Jumlah Sampel Uji : 3 (tiga)  
 Tanggal Penerimaan : 9 Juli 2021  
 Tanggal Pengujian : 3 - 20 Agustus 2021

No.	Parameter Uji	Satuan	Z1	Z2	Z3	Metode
			Titik 21.528	Titik 21.529	Titik 21.530	
1	KTK*	cmol(+)kg <sup>-1</sup>	36.93	50.17	26.50	Destilasi IK 5.4.f

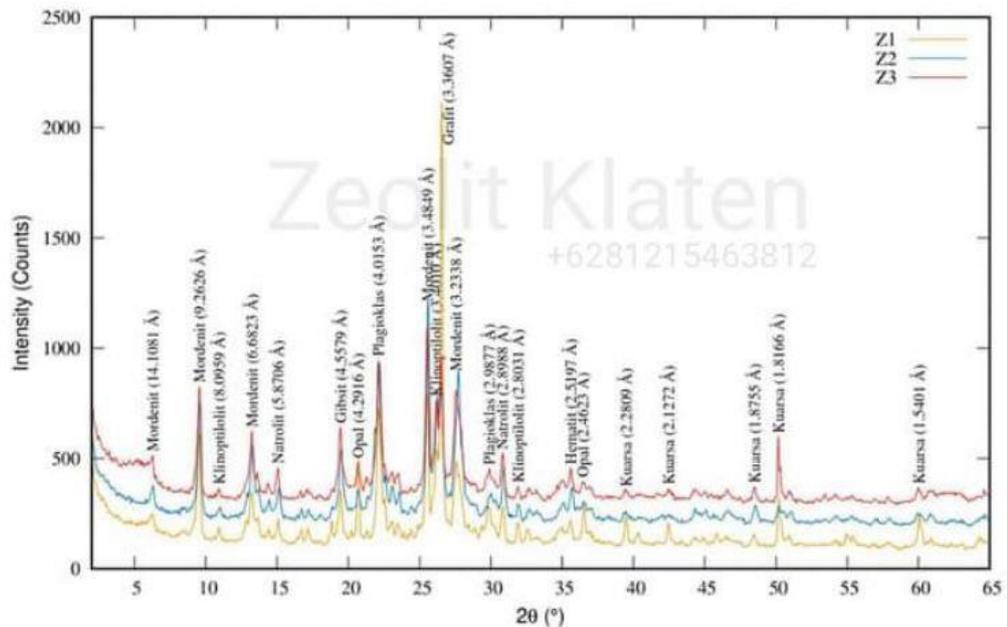
Hasil analisis ini hanya berlaku untuk sampel yang dimaksud

" Tidak dibenarkan mengandakan sebagian seluruh isi hasil analisis ini, tanpa izin Laboratorium BPTP Yogyakarta dan pemilik hasil analisis"

Yogyakarta, 23 Agustus 2021  
 Deputy Manajer Teknis.

*(Signature)*  
 Widada, A.Md  
 NIP. 196807121990031001

Keterangan : \* Parameter terakreditasi



Sampel Z1				
No	Element		Concentration	Abs. Error
11	Na <sub>2</sub> O	Sodiumoxide	0,992%	0.018
12	MgO	Magnesiumoxide	0,5970%	0.0066
13	Al <sub>2</sub> O <sub>3</sub>	Aluminumoxide	8,409%	0.011
14	SiO <sub>2</sub>	Siliconoxide	61,92%	0.02
15	P <sub>2</sub> O <sub>5</sub>	Phosphorusoxide	0,04677%	0.00071
16	SO <sub>3</sub>	Sulfideoxide	0,00786%	0.00012
17	Cl	Chlorine	0,00787%	0.00009
19	K <sub>2</sub> O	Potassiumoxide	1,072%	0.002
20	CaO	Calciumoxide	4,712%	0.003
21	Fe <sub>2</sub> O <sub>3</sub>	Ironoxide	3,011%	0.004
22	MnO	Manganesoxide	1284 µg/g	1
23	Sc	Scandium	80.8 µg/g	3.1
24	Ti	Titanium	1830 µg/g	3
25	V	Vanadium	38.9 µg/g	1
26	Cr	Chromium	6.2 µg/g	0.3
27	Co	Cobalt	< 0.1 µg/g	-
28	Ni	Nickel	14.5 µg/g	0.9
29	Cu	Copper	16 µg/g	0.8
30	Zn	Zinc	42.1 µg/g	0.7
31	Ga	Gallium	13.2 µg/g	0.3
32	Ge	Germanium	0.4 µg/g	0.2
33	As	Arsenic	0.8 µg/g	0.2
34	Se	Selenium	0.2µg/g	0.1
35	Br	Bromine	< 0.1 µg/g	-
37	Rb	Rubidium	29.9 µg/g	0.2
38	Sr	Strontium	677.1 µg/g	0.8
39	Y	Yttrium	15.9 µg/g	0.2
40	Zr	Zirconium	102.9 µg/g	0.5
41	Nb	Niobium	< 0.1 µg/g	-
42	Mo	Molybdenum	< 0.1 µg/g	-
44	Ru	Ruthenium	0.8 µg/g	0.2
45	Rh	Rhodium	0.6 µg/g	0.1
46	Pd	Palladium	< 0.4 µg/g	-
47	Ag	Silver	2.4 µg/g	0.1
48	Cd	Cadmium	0.5 µg/g	0.2
49	In	Indium	0.8 µg/g	0.2
50	Sn	Tin	4.1 µg/g	0.5
51	Sb	Antimony	5.1 µg/g	0.6
52	Te	Tellurium	≈ 18.8 µg/g	1.7
53	I	Iodine	12.9 µg/g	1
55	Cs	Cesium	27.1 µg/g	2
56	Ba	Barium	133.2 µg/g	2.3
57	La	Lanthanium	52.5 µg/g	4.1
58	Ce	Cerium	31.8 µg/g	2.9
59	Pr	Praseodymium	5.3 µg/g	2
60	Nd	Neodymium	< 0.5 µg/g	-
62	Sm	Samarium	< 0.1 µg/g	-
70	Yb	Ytterbium	< 4.2 µg/g	-
72	Hf	Hafnium	2.7 µg/g	0.8
73	Ta	Tantalum	< 0.5 µg/g	-
74	W	Tungsten	< 0.1 µg/g	-
79	Au	Gold	< 0.2 µg/g	-
80	Hg	Mercury	< 0.3 µg/g	-
81	Tl	Thallium	< 0.1 µg/g	-
82	Pb	Lead	4.5 µg/g	0.4
83	Bi	Bismuth	< 0.4 µg/g	-
90	Th	Thorium	0.4 µg/g	0.2
92	U	Uranium	0.9 µg/g	0.4

XRDZeolite-results

**ZEOLIT KLATEN**

081215463812

<b>SAMPLE</b>	<b>Z1</b>	<b>Z2</b>	<b>Z3</b>
<b>Klinoptilolit</b>	<b>40,46</b>	<b>37,35</b>	<b>33,23</b>
<b>Mordenit</b>	<b>11,46</b>	<b>14,29</b>	<b>14,27</b>
Ortoklas	13,79	11,01	16,77
Plagioklas	6,66	15,20	10,95
Opal	5,29	11,53	13,48
Grafit	6,74	4,31	2,98
Natrolit	3,23	3,20	3,57
Kuarsa	8,95	-	-
Hematit	1,74	1,60	2,50
Gibsit	1,68	1,51	2,25
Total	100,00	100,00	100,00